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**Research Article** 

# ANDROGRAPHOLIDE AND ITS DERIVATIVE - A STORY OF ANTIMALARIAL DRUG DESIGN AND SYNTHESIS

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# ABSTRACT

**Objective:** Andrographolide was found to show moderate antimalarial activity against chloroquine-resistant strain of *Plasmodium falciparum* (PF). It thus becomes an interesting lead for new antimalarial drugs. This study describes a molecular docking of andrographolide and its derivative into the best PF geranylgeranyl pyrophosphate synthase (PFGGPPS) model.

**Methods:** A comparative modeling of PFGGPPS based on a crystal structure of *Plasmodium vivax* GGPPS was optimized and conducted. This model was considered suitable for molecular docking. Partition coefficient of andrographolide was determined to assist its derivative design based on hydrophobicity property. Synthesis of the antimalarial drug was scaled up to 5 mm and identified by<sup>13</sup> C- and <sup>1</sup>H-nuclear magnetic resonance (NMR) spectroscopy.

**Results:** The optimal comparative modeling of PFGGPPS was conducted on chain B (3PH7 chain B). The calculated coefficient partition of andrographolide's derivative was higher (+1.89), compared to that of andrographolide of +1.62. The NMR data of second and third synthesis experiments were consistent at the 5-mmol scale.

**Conclusions:** On the molecular docking of andrographolide into the model, an antimalarial andrographolide derivative design that is more hydrophobic than andrographolide was proposed since the stronger hydrophobicity property of drug, the better of its activity of the drug. Synthesis of this derivative with a simple and green procedure was found to be reproducible.

**Keywords:** *Plasmodium falciparum*, Geranylgeranyl pyrophosphate synthase, Andrographolide, Comparative modeling, Molecular docking, Drug design and synthesis.

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# INTRODUCTION

*Plasmodium falciparum* (PF), a microorganism that belongs to the phylum *Apicomplexa* [1], is the most common cause of malaria in Indonesia [2]. The first-line treatment for uncomplicated PF malaria in the country is the artesunate-amodiaquine combination [2]. However, a failure rate of more than 10%, associated with drug resistance, was reported for this treatment in Indonesia for the period 2000-2007 [2]. This report presents the basis for a new antimalarial drug development in Indonesia. Geranylgeranyl pyrophosphate synthase (GGPPS) is a key enzyme in PF isoprenoid biosynthesis [3]. It catalyzes the synthesis of GGPP,<sup>3</sup> as well as farnesyl pyrophosphate [3]. In the presence of Mg<sup>2+</sup> cofactors [4]. It is recommended as an antimalarial drug target due to its amenability to high-throughput screening, druggability, and essentiality [5].

Andrographolide (Fig. 1) is the major secondary metabolite in an Indonesian herb, *Sambiloto (Andrographis paniculata)* [6]. It was found to have a moderate antimalarial activity against a chloroquine-resistant strain of PF [7], making it an interesting lead for new antimalarial drugs. Since it is biosynthesized from and similar to GGPP [8], it is reasonable to hypothesize that andrographolide exerts its antimalarial activity by inhibition of GGPPS. To establish this hypothesis by molecular docking, a crystal structure of PFGGPPS is needed, but this is not yet available.

Meanwhile, high-resolution GGPP-binding crystal structures of *Plasmodium vivax* GGPPS (PVGGPPS) are available (Protein Data Bank [PDB] 3CC9 [9] and 3PH7 [4]), based on a sequence available at UniProt [10] (A5K4U6 [11]). A basic local alignment search tool search on UniProt for this sequence gave a PFGGPPS sequence (Q86GK8 [12])

that was evidenced from transcript level. A >70% similarity between the two sequences was revealed in our preliminary sequence alignment by ClustalX 2.1 [13]. Therefore, it is feasible to make a comparative modeling of PFGGPPS using one the PVGGPPS crystal structures as the template.

In this paper, we report the aforementioned comparative modeling. Then, we describe a molecular docking of andrographolide into the best PFGGPPS model. Based on the results of this molecular docking, we propose an antimalarial andrographolide derivative design (Fig. 1), and we report our attempt to synthesize it.

#### METHODS

#### **Comparative modeling of PFGGPPS**

The comparative modeling was conducted by Modeller  $9v^{16}$  interfaced by EasyModeller 4.0 [16]. A sequence of PFGGPPS downloaded from UniProt [10] as a FASTA file (Q86GK8 [12]) was employed as the query. A 2.5 Å-resolution GGPP-binding crystal structure of PVGGPPS downloaded from the Research Collaboratory for Structural Bioinformatics (RCSB). PDB [17] as a PDB file (PDB 3PH7) [14] was selected as the template. During the comparative modeling, heteroatoms were not included, and loops were not refined. Instead, the model with the lowest discrete optimized protein energy (DOPE) score was optimized by default parameters.

## Preparation of PFGGPPS for molecular docking

The optimized model of PFGGPPS from the comparative modeling was prepared for molecular docking by OEDocking 3.0.1 [18]. Detection of its active site was carried out using a molecular method, which was guided by Imprint Pocket [19]. The largest cavity indicated by Imprint Pocket was selected as the active site. Detection sensitivity was adjusted manually to ensure that all active site residues were covered. The molecular docking grid box was defined as  $26.33 \times 43.00 \times 30.33$ Å<sup>3</sup> = 34,347 Å<sup>3</sup>. Site shape potential was balanced (extended toward solvent and protein), and outer contour was set to 1,747 Å<sup>3</sup>. No constraint was applied.

#### Preparation of andrographolide and its derivative for molecular docking

A crystal structure of andrographolide (1) in CIF format [20] was used as a starting conformation of andrographolide. The format was converted into PDB by PyMOL 1.3 [21], and then this was converted into OpenEye native format (oeb) by VIDA 4.2.1 [22]. The structure of the andrographolide's derivative (2) was built on this crystal structure by VIDA 4.2.1 [22]. Before molecular docking, every ligand was fixed to pKa = 7.0 and their tautomers assigned by QUACPAC 1.6.3.1 [23,24]. Then, the derivative was filtered for drug-likenesses and its conformations generated by OMEGA 2.5.1.4 [25] using default parameters.

#### Molecular docking of andrographolide and its derivative

Molecular docking of the prepared ligands was conducted by OEDocking 3.0.1 [18] *in vacuo* using default parameters. Every best docking pose was then subjected to optimization by SZYBKI 1.8.0.2 [26], considering a minimum distance of 2 Å between PFGGPPS active site residues and the ligand. In this optimization, protein-ligand interaction energies were calculated using Merck molecular force field (MMFF) van der Waals potential, in which MMFF Coulomb was digitized on the grid.

# Calculation of partition coefficient of andrographolide and its derivative

A calculation of partition coefficient (log P) of andrographolide was conducted by ACD/Labs 12.01 [15] to assist the design of 2.

#### Synthesis of andrographolide derivative

Andrographolide (≥98%) was purchased from Xi'an App Chem-Bio(Tech) Co., Ltd., P.R. China, and would expire on May 24, 2015. It appears as a white powder that dissolves in chloroform, ethyl acetate, and methanol [27,28], and it decomposes at 70°C [29].

To modify 1 into 2, andrographolide (1 mmol) was dissolved in methanol and reacted with anhydrous sodium carbonate (1 mmol) at 50°C for 2 hrs, a procedure adapted from Pandeti *et al.* [30] The reaction was monitored by thin-layer chromatography every hour. The desired product was purified by solvent-solvent extraction (chloroformwater 1:1), which was followed by column chromatography (eluent: 100% ethyl acetate) and identified by <sup>13</sup>C- and <sup>1</sup>H-nuclear magnetic resonance (NMR) spectroscopy. The same experiment was repeated to assess reproducibility. Following a good confirmation, the procedure

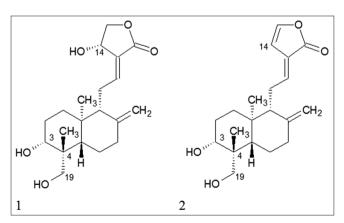


Fig. 1: Andrographolide (1) [14] and its proposed derivative (2) (drawn by ACD/Labs 12.01) [15]

was scaled up to 5 mmol. Only at this point was the reaction yield calculated.

# RESULTS AND DISCUSSION

#### **Comparative modeling of PFGGPPS**

Before the comparative EasyModeller, Modeler, aligned sequences of the template and query. For chains A to D of the template, the alignment scores were 265257.6250, 273733.8750, 252205.1562, and 260121.5312, respectively. Since chain B of the template gained the highest score for the alignment, and no gap presented at predicted secondary structures of the query (Fig. 2), the comparative modeling was conducted on chain B.

The comparative modeling produced seven models with the following DOPE scores: -45876.36719, -45724.93750, -45170.59375, -45537.06250, -45908.73047, -45727.52344, and -45883.46484. Model 5 received the lowest DOPE score, so it was optimized. After optimization of this model, a Ramachandran plot revealed only a few residues outside the allowed regions (Fig. 3). One of these residues was part of the PFGGPPS active site, according to Imprint Pocket (Fig. 2). Therefore, this PFGGPPS model was considered suitable for molecular docking. A comparison between the PVGGPPS crystal structure and PFGGPPS model is depicted in Fig. 4.

#### Molecular docking of andrographolide and its derivative

The active site of PVGGPPS consists of a hydrophobic anchor ( $L^{118}$ ,  $Q^{119}$ ,  $A^{120}$ ,  $F^{122}$ ,  $V^{156}$ ,  $Y^{160}$ ,  $T^{188}$ , and  $I^{192}$ ; numbering differs 20 amino acids more than sequence numbering) and a catalytic pocket containing the first aspartate-rich motif (FARM; <sup>126</sup>DDIMD<sup>130</sup>) and second aspartate-rich motif (SARM; <sup>287</sup>DDYID<sup>291</sup>) [4]. GGPP positions itself in the PVGGPPS

	1 10	20	30	40
3PH7B		LAFFRNMYDK		
Q86GK8	MENEQNNQDSENG	LDYFRSMYDR	<u>YRDVFI</u> <u>NHI N</u>	DYVLEDD
	50	60	70	80
3PH7B	IKEHISKYYKLLFDY			
Q86GK8	IKI I I SKYYKLLFD	<u>YNCLGGKMNRC</u>	<u>HLVILIYEY</u> VK	NRDI
	90	100	110	12
3PH7B	NS SEWEKAACLAW			
Q86GK8	NCNEWEKVACI AWCIEILOAS FLVADDIMDKGET RRNKHC			
	130	140	150	160
3PH7B	WYLLKDVETKNAV	NDVLLLYNSIY	KLI EI YLRN	ESCYVDV
Q86GK8	WILLKDVE/KNAM	ND VFLL YNAIYI	KLLDVYLRND	NCYLDL
	170	180	190	200
3PH7B	IATFRDATLKTII GO	HLDTNIFSDKY	SDAHREIDVN	ININV
Q86GK8	ITS FREATLKTIVGO	HLDTNIF <u>SDKY</u>	<u>SHI D</u> KDIDVN	NINI
	210	220	230	240
3PH7B	PQP-V IDINMI NFG			
Q86GK8	SOENKININMLNFKV	<u>YONI I IHK</u> TAY	YYSFFLPIVCG	MOMG
	250	260	270	280
3PH7B	GI-DNLIYKK I ED	ISMLMGEYFQ	I HDDYLDI FG	DSTKTG
Q86GK8	GISLDNLLYKKVENIAI LMGEYFOVHDDYI DIFGDSKKTG			
	290	300	310	320
3PH7B	KVGSDIQNNKLTWF			
Q86GK8	KVG <u>SDIONN</u> KLTWPL	. <u>IKAFEL</u> C <u>SOPE</u>	KEDII RNYGA	D <u>NVT</u>
	330	340	350	360
3PH7B	CVKVIDSLYEQYKI			
Q86GK8	CI KF INDIYEHYNIR	DHYVEYEKKO	KMKILEAINO	LHHEG
	370			
3PH7B	IEYVLKYLLEILFTG	v		

Fig. 2: Sequence alignment of the template (3PH7 chain B) and query (Q86GK8). Gaps in the alignment are indicated by red dash; predicted alpha helix structures for the query are constructed by underlined amino acids; no beta structure was predicted for the query by modeller; the active *Plasmodium falciparum* geranylgeranyl pyrophosphate synthase site, according to Imprint Pocket, is constructed by italicized amino acids; outliers in the Ramachandran plot are colored red in Fig. 3 active site such that its hydrophobic part is held by the anchor while its pyrophosphate group interacts with the catalytic pocket [4]. In our docking pose of andrographolide, the hydroxyl groups at C-3 and C-19 pointed toward the SARM of PFGGPPS, while its lactone group pointed toward the hydrophobic anchor (Fig. 5).

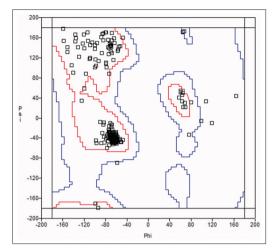


Fig. 3: Ramachandran plot of the optimized *Plasmodium falciparum* geranylgeranyl pyrophosphate synthase model 5

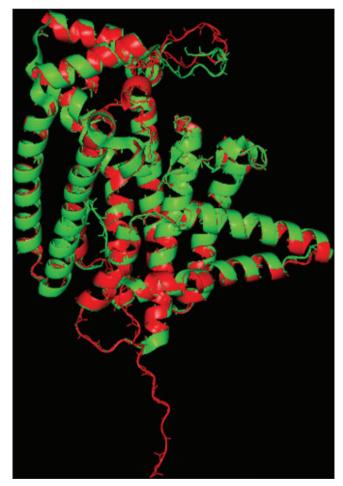


Fig. 4: Comparison between the *Plasmodium vivax* geranylgeranyl pyrophosphate synthase crystal structure (green) and our *Plasmodium falciparum* geranylgeranyl pyrophosphate synthase model (red) (image generated by PyMOL 1.3) [21]

The overall energy of the interaction between andrographolide and PFGGPPS was -70.81114 kcal/mol (comprising electrostatic and van der Waals interactions of -58.43179 and -12.37936 kcal/mol, respectively). Stronger electrostatic interactions should be expected in the presence of Mg<sup>2+</sup> cofactors and water molecules. This finding suggests that the andrographolide hydroxyl groups at C-3 and C-19 were essential for hydrophilic interactions, while its lactone group could be modified to become more hydrophobic.

To examine this suggestion, we proposed a design for an andrographolide derivative (2) which was more hydrophobic than andrographolide at its lactone group. On docking into PFGGPPS, this derivative gave a highly similar position and orientation (Fig. 5). Its affinity toward PFGGPPS was lost slightly, as indicated by a decrease in the overall ligand-protein interaction energy (down by about an 8<sup>th</sup>, to –61.90828 kcal/mol). However, the van der Waals contribution in the interaction energy increased by approximately one-third, to –16.40830 kcal/mol, suggesting that the pre-modified lactone group at two bound more tightly to the hydrophobic anchor of PFGGPPS.

Since an antimalarial drug should penetrate the membrane of PF cell and work within the cell, its activity is part determined by its hydrophobicity, which is reflected by its partition coefficient (log P): The higher its log P (until +5.00, according to Lipinski's Rule of Five) [31], the better it penetrates the membrane and the higher is its activity. The calculated log P of 2 is +1.89, compared to that of andrographolide of +1.62. Therefore, 2 is expected to penetrate the membrane better. This pharmacokinetic profile could compensate the previous pharmacodynamic one.

#### Synthesis of the andrographolide derivative

The above procedure was selected because it is considered simple and sufficiently green, so it is suitable for drug synthesis. Below is the result of the <sup>13</sup>C- and <sup>1</sup>H-NMR for the first experiment:

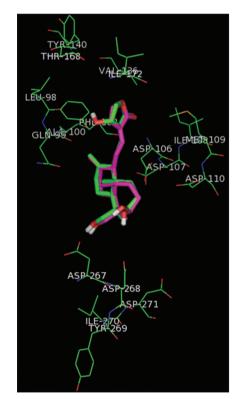


Fig. 5: Docking pose of andrographolide (green sticks) and derivative (pink sticks) in *Plasmodium falciparum* geranylgeranyl pyrophosphate synthase. The first aspartate-rich motif is on the right, and the second aspartate-rich motif is at the bottom; on the left is the hydrophobic anchor of PFGGPPS active site (image generated by PyMOL 1.3) [21]

<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 198.53, 169.88, 167.27, 148.10, 146.89, 146.69, 142.65, 142.58, 124.22, 124.15, 123.44, 108.59, 108.43, 108.15, 102.34, 80.19, 77.46, 77.34, 77.14, 76.82, 64.15, 56.72, 56.00, 55.70, 55.10, 55.05, 52.16, 42.64, 42.02, 38.84, 38.76, 37.78, 37.68, 36.94, 36.88, 32.77, 32.73, 28.04, 25.63, 24.38, 23.68, 22.81, 15.22, 15.18.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 9.68, 9.59, 9.56, 7.50, 7.24, 7.14, 6.97, 6.89, 6.82, 6.78, 6.75, 6.62, 6.42, 6.09, 6.07, 6.02, 5.97, 5.42, 5.31, 5.04, 4.91, 4.77, 4.76, 4.71, 4.54, 4.45, 4.38, 4.33, 4.30, 4.12, 4.09, 3.66, 3.45, 3.40, 3.38, 3.24, 3.22, 2.95, 2.91, 2.74, 2.64, 2.60, 2.47, 2.35, 2.32, 2.22, 2.12, 1.91, 1.89, 1.73, 1.66, 1.49, 1.45, 1.41, 1.35, 1.32, 1.26, 1.22, 1.16, 0.81, 0.74, 0.60, 0.44, -0.00.

These NMR data were consistent for the second and third experiments, suggesting a high reproducibility for the procedure. At the 5-mmol scale, the reaction yield was 60.94%.

The <sup>13</sup>C-NMR peak near 200 ppm suggests that the andrographolide hydroxyl group at C-19 may have been oxidized by sodium carbonate into an aldehyde. This by-product would be anticipated, since the andrographolide hydroxyl group at C-19 is also vulnerable and selective protection had not been applied. We have to bear in mind also that protection requires more steps and may not be sufficiently green for drug synthesis [32]. The presence of this by-product should be investigated further by analytical and preparative high-performance liquid chromatography and liquid chromatography-mass spectroscopy.

#### CONCLUSION

A comparative modeling of PFGGPPS based on a PVGGPPS crystal structure (3PH7 chain B) was conducted, and the model was considered suitable for molecular docking. On molecular docking of andrographolide into the model, the design of an antimalarial andrographolide derivative, which is more hydrophobic than andrographolide, was proposed. Synthesis of this with a simple and green procedure was found to be reproducible at a 5-mmol scale.

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