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**Original Article** 

# DEVELOPMENT AND VALIDATION OF NOVEL RP-HPLC METHOD FOR SIMULTANEOUS ESTIMATION OF GABAPENTIN AND AMITRIPTYLINE HYDROCHLORIDE IN BULK AND PHARMACEUTICAL DOSAGE FORMS

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# ABSTRACT

**Objective**: To develop a novel, accurate, precise and linear reverse phase high performance liquid chromatographic (RP-HPLC) method for simultaneous quantitative estimation of gabapentin and amitriptyline hydrochloride in gabantip-at tablet and validate as per international conference on harmonization (ICH) guidelines and to perform the force degradation studies using the developed method.

**Methods**: In the present work, good chromatographic separation was achieved isocratically using a shim-pack HPLC C18 column (4.6 x 250 mm, 5µm) and mobile phase consisting of 0.05 M potassium dihydrogen orthophosphate pH 2.1 adjusted with orthophosphoric acid and acetonitrile in the ratio (55:45), at flow rate 1 ml/min and column temperature (25 °C). The effluents obtained were monitored at 221 nm with the UV-visible detector.

**Results**: The retention time of gabapentin and amitriptyline hydrochloride was found to be 1.959 min and 4.221 min respectively. The linearity of gabapentin was found in the range of 720-1680 ppm and that for amitriptyline hydrochloride was found to be 24-56 ppm. The correlation coefficient for gabapentin and amitriptyline hydrochloride were 0.999 and 0.9963 respectively. The high recovery values (98%-101%) indicate a satisfactory accuracy. The low percent relative standard deviation (% RSD) values in the precision study reveals that the method is precise.

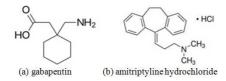
**Conclusion**: The developed method is novel, simple, precise, rapid, accurate and reproducible for simultaneous estimation of gabapentin and amitriptyline hydrochloride tablet dosage form. Hence the proposed method may find practical applications as a quality-control tool in the simultaneous analysis of the two drugs in combined dosage forms in quality-control laboratories.

Keywords: Gabapentin, Amitriptyline hydrochloride, RP-HPLC, Analysis, Validation, ICH

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# INTRODUCTION

The IUPAC name of the gabapentin and amitriptyline hydrochloride is 2-[1-(aminomethyl) cyclohexyl] acetic acid and 3-(5,6dihydrodibenzo[2,1-b: 2',1'-f] 7-annulen-11-ylidene)-N,N-dimethylpropan-1-amine respectively, with molecular formula  $C_9H_{17}NO_2$  and  $C_{20}H_{23}N$ . HCl respectively and molecular weight 207.70 and 313.87 respectively [1, 2]. Gabapentin is freely soluble in water, alkaline and acidic solution, sparingly soluble in methanol [1]. Amitriptyline hydrochloride is freely soluble in water and in alcohol [2]. The molecular structure of the drugs is given in fig. 1.



# Fig. 1: Chemical structure of (a) gabapentin and (b) amitriptyline hydrochloride [1, 2]

Gabapentin is an antiepileptic medication. It decreases pain and seizures by modulating calcium channel activity of the nerve cells [1] and amitriptyline is used in the treatment of depression and neuropathic pain. Amitriptyline increases the levels of chemical messengers in the brain that help in regulating the mood and treat depression [2, 3].

Combination therapy of gabapentin and amitriptyline hydrochloride is used for the treatment of neuropathic pain, depression and seizures. Combination of gabapentin and nortriptyline is also available in the market but the fact is amitriptyline is more effective in the treatment of endogenous depressive illness then nortriptyline [4] and it should remain in its position as the gold-standard antidepressant [5]. Literature survey reveals that various analytical and HPLC methods for estimation of gabapentin and amitriptyline hydrochloride were reported alone and in combination with other drugs in bulk, dosage forms, human plasma and in urine sample [1, 2, 6-16]. Also, HPLC methods have been reported for gabapentin and nortriptyline hydrochloride combination [17, 18] but to the best of our knowledge, there is no such reported HPLC analysis method for simultaneous estimation of gabapentin and amitriptyline hydrochloride combination. Hence, the aim of the present study was to develop a rapid and precise RP-HPLC method for the simultaneous estimation of gabapentin and amitriptyline hydrochloride in the combined dosage form and validate the developed method in accordance with ICH guidelines and also to perform the force degradation studies using the developed method. This novel validated method has applicability in the industry as well as academia.

# MATERIALS AND METHODS

Gabapentin (99% potency) and amitriptyline hydrochloride (99% potency) were purchased from Chem dyes Corporation, Rajkot-Gujarat. A commercial preparation gabantip-at tablet 'Sun Pharmaceutical Industries Ltd' containing gabapentin 300 mg and amitriptyline hydrochloride 10 mg used for analysis was procured from local market. HPLC grade solvents were purchased from Thomas Baker. RP-HPLC shimadzu (LC 2030) model with "Lab Solution" software was employed in this method. Analytical column used for the separation of analytes was shim-pack HPLC C18 (250 X 4.6 mm, 5  $\mu$ m).

#### Methods

#### Selection of wavelength

The suitable wavelength for the HPLC analysis was determined by recording UV spectrums in the range of 190-380 nm for individual drug solutions of amitriptyline hydrochloride and gabapentin then

overlapped. UV overlain spectra of both gabapentin and amitriptyline hydrochloride showed that both the drugs absorb

appreciably at 221 nm and hence 221 nm was taken as a detection wavelength for HPLC analysis (fig. 2).

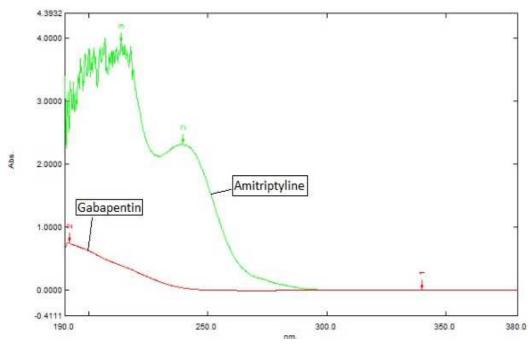


Fig. 2: UV overlap spectrum of amitriptyline hydrochloride and gabapentin

#### **Chromatographic conditions**

The method was developed using reverse phase, shim-pack HPLC C18 column (250 X 4.6 mm, 5  $\mu$ m). The runtime was of 7 min. The mobile phase used was 0.05 M potassium dihydrogen orthophosphate pH 2.1 adjusted with orthophosphoric acid and acetonitrile in the ratio (55:45) at a flow rate of 1.0 ml/min, column temperature (25 °C) and a detection wavelength of 221 nm using a UV-visible detector.

# Preparation of 0.05 M phosphate buffer (pH 2.1)

About 6.80 g of potassium dihydrogen orthophosphate was accurately weighed and dissolved in 950 ml of water. The pHwas adjusted to 2.1 with orthophosphoric acid and the volume was made up to 1000 ml in a volumetric flask. The solution was then filtered using 0.45  $\mu$  membrane filter.

#### **Preparation of standard solution**

100 mg of amitriptyline hydrochloride and 300 mg of gabapentin standard were accurately weighed and transferred into 100 ml volumetric flask respectively. About 70 ml solvent was added, sonicated to dissolve and diluted up to the mark using a solvent (1000 ppm of amitriptyline hydrochloride and 3000 ppm of gabapentin). The final concentration of amitriptyline hydrochloride and gabapentin were made to 40 ppm and 1200 ppm respectively by suitable dilutions.

# Preparation of sample solution

10 tablets were weighed and powdered. The quantity of powder equivalent to 10 mg of amitriptyline hydrochloride and 300 mg of gabapentin was transferred into a 100-ml volumetric flask. About 70 ml solvent was added and the solution was sonicated for 30 min with intermittent shaking. The volume was made up using the solvent, mixed and filtered through 0.45  $\mu$  PVDF filter. Final concentrations of amitriptyline hydrochloride and gabapentin were made to 40 ppm and 1200 ppm respectively with suitable dilution.

# Forced degradation studies

Forced degradation is the process of subjecting drug compounds to extreme chemical and environmental conditions to determine product breakdown levels and preliminary degradation kinetics, and to identify potential degradation products. They are used to facilitate the development of analytical methodology, to gain a better understanding of active pharmaceutical ingredient and drug product stability, and to provide information about degradation pathways and degradation products. The study involves acid and alkali hydrolysis wherein sample solution was treated with 0.1 N HCl and 0.1 N NaOH respectively at room temperature for 2 h. Oxidative degradation studies involved 30% v/v H<sub>2</sub>O<sub>2</sub> treatment of sample solution at room temperature for 2 h. The samples were placed in hot air oven at 105 °C for 1 h to study thermal degradation. For photolytic stress studies, samples were exposed in photostability chamber for 5 h. The sample was placed in a humidity chamber at 25 °C and 80 % RH (relative humidity) for 24 h to study degradation by humidity [19, 20].

#### **RESULTS AND DISCUSSION**

#### Method development

A reverse phase HPLC method was developed keeping in mind the system suitability parameters i.e. resolution factor between peaks, tailing factor, a number of theoretical plates, runtime and the costeffectiveness. The developed optimized method resulted in the elution of gabapentin at 1.9 min and amitriptyline hydrochloride at 4.2 min. Fig. 3, 4 and 5 represent chromatograms of the blank solution, gabapentin standard solution and amitriptyline hydrochloride standard solution for the solution and amitriptyline hydrochloride standard solution and amitriptyline hydrochloride standard solution for the solution and amitriptyline hydrochloride standard solution for the solution and amitriptyline hydrochloride standard solution for the solution for t

The total run time was 7 min. System suitability tests are an integral part of method development and are used to ensure adequate performance of the chromatographic system. Retention time, a number of theoretical plates, peak resolution and peak tailing factor were evaluated for six replicate injections of the standard working concentration. The results given in table 1 were within acceptable limits [21].

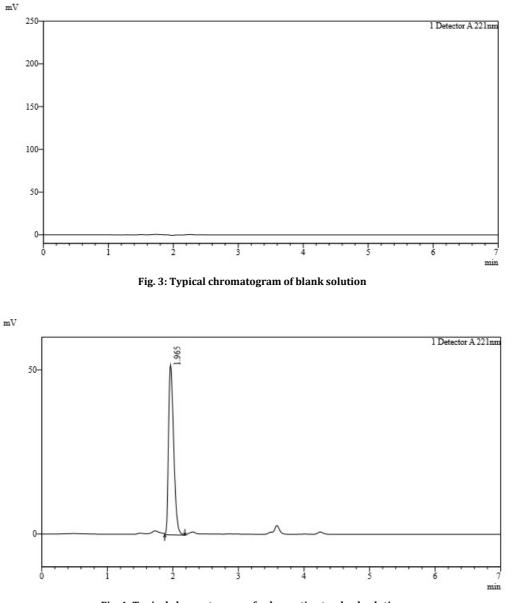


Fig. 4: Typical chromatogram of gabapentin standard solution

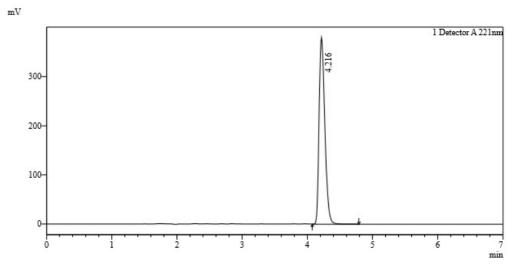


Fig. 5: Typical chromatogram of amitriptyline hydrochloride standard solution

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#### Table 1: Results of system suitability studies

Parameters	Acceptance limits	Amitriptyline hydrochloride	Gabapentin
Retention time	-	4.221 min	1.959 min
Resolution factor	Not less than 2	13.575	
Number of Theoretical plate	Not less than 2000	9450	2415
Tailing factor	Not more than 2	1.355	1.508

In order to test the applicability of the developed method to a commercial formulation, 'gabantip-at' tablets were chromatographed at working concentration and it is shown in fig. 6. The sample peaks were identified by comparing the relative retention times with standard drugs (fig 4-5). System suitability parameters were within the acceptance limits, ideal for the

chromatographed sample. Integration of separated peak area was done and each drug concentration was determined by using the peak area concentration relationship obtained in the standardization step. The protocol affords reproducible quantification of the two drugs with an error less than 10 %, which is the standard level in any pharmaceutical quality control.

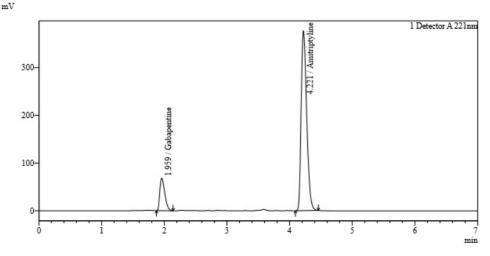


Fig. 6: Typical chromatogram of the sample

# Method validation

Validation of the analytical method is the process that establishes by laboratory studies in which the performance characteristics of the method meet the requirements for the intended analytical application. HPLC method developed was validated according to ICH guidelines [22] for validation of analytical procedures. The method was validated for the parameters like linearity, accuracy, system precision, method precision, robustness, limit of detection and limit of quantitation.

# Specificity

Fig. 3-6 for a blank solution, standard drug solutions and sample chromatogram reveal that the peaks obtained in the standard solutions and sample solution at working concentrations are only because of the drugs as blank has no peak at the retention time of gabapentin and amitriptyline hydrochloride. Accordingly, it can be concluded that the method developed is said to be specific [23].

#### Precision

#### System precision

Six replicate injections of the standard solutions at working concentration showed percent relative standard deviation (% RSD) less than 2 concerning peak area for both the drugs, which indicates the acceptable reproducibility and thereby the precision of the system [24, 25]. System precision results are tabulated in table 2.

#### Method precision

Method precision was determined by performing the assay of the sample under the test of repeatability (intraday precision) at working concentration. Six injections of the sample from the same homogeneous mixture at working concentration showed % RSD less than 2 concerning % assay for both the drugs which indicate that the method developed is method precise by the test of repeatability [24, 25] and hence can be understood that the method gives consistently reproducible results (table 3).

#### **Table 2: System precision results**

S. No.	Peak area of amitriptyline hydrochloride (40 ppm)	Peak area of gabapentin (1200 ppm)
1	2325911	382152
2	2331400	382984
3	2341426	386194
4	2338850	385431
5	2343853	387636
6	2341335	386683
Average	2337129	385180
SD	6971	2162
%RSD	0.30	0.56

#SD: Standard deviation, #% RSD: Percent relative standard deviation

#### **Table 3: Method precision results**

S. No.	Amitriptyline hydrochloride % assay	Gabapentin % assay	
1	99.31	99.70	
2	99.98	99.20	
3	100.92	100.20	
4	99.48	100.20	
5	100.35	99.70	
6	99.58	100.30	
Average	100.10	99.90	
SD	8.40	4.30	
%RSD	0.84	0.43	

#SD: Standard deviation, #%RSD: Percent relative standard deviation

# Linearity

Standard solutions of amitriptyline hydrochloride and gabapentin at different concentrations level (60%, 80%, 100%, 120%, and 140%) were prepared in triplicates. Calibration curves were constructed by plotting the concentration level versus corresponding peak areas for both the drugs. The results show an excellent correlation between peak areas and concentrations level within the tested concentration range of 24-56 ppm for amitriptyline hydrochloride and 720-1680 ppm for gabapentin (table 4). The correlation coefficients were greater than 0.99 for both the drugs, which meet the method validation acceptance criteria [24, 25] and hence the method is said to be linear for both the drugs (fig. 7-9).

Table	4: Data	for lin	earity	studies
Tuble	I. Dutu	IOI IIII	curity	studies

% level	Amitriptyline hydrochloride	Amitriptyline hydrochloride		
	Concentration in ppm	Peak area	Concentration in ppm	Peak area
60	24	1420116	720	242796
80	32	1951615	960	311701
100	40	2325911	1200	382152
120	48	2834560	1440	445676
140	56	3389963	1680	514464
Slope	60283		282.36	
Y intercept	-26887		40557	
Correlation Coefficient	0.9963		0.999	

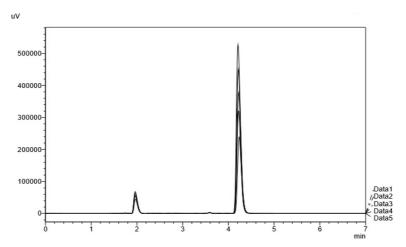


Fig. 7: Chromatogram of gabapentin and amitriptyline hydrochloride (five different concentration overlapped for linearity study)

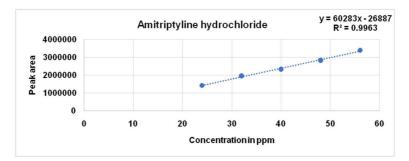
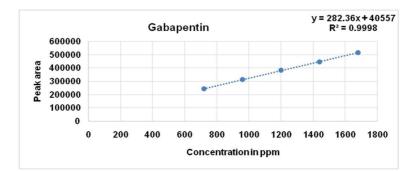
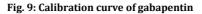


Fig. 8: Calibration curve of amitriptyline hydrochloride





#### Accuracy

Accuracy was determined by means of recovery experiments, by the determination of % mean recovery of both the drugs in the formulation at three different levels (80-120%). At each level, three

determinations were performed. Percent mean recovery was calculated as shown in table 5 and 6. The accepted limits of mean recovery are 98%-102% and all observed data were within the required range, which indicates good recovery values, affirming the accuracy of the method developed [24, 25].

#### Table 5: Accuracy data for gabapentin

Level (%) (n=3) (gabapentin)	Sample [ppm]	Standard added [ppm]	Actual amount [ppm]	Total area found [average] n=3	Amount recovered [ppm]	% recovery
80	600	480	1080	350140	1096.41	101.51
100	600	600	1200	382152	1209.78	100.81
120	600	720	1320	412115	1315.90	99.68

#n: Number of injections

#### Table 6: Accuracy data for amitriptyline hydrochloride

Level (%) (n=3) (amitriptyline hydrochloride)	Sample (ppm)	Standard added (ppm)	Actual amount [ppm]	Total area found [average] n=3	Amount recovered [ppm]	% recovery
80	20	16	36	2166897	36.39	101.08
100	20	20	40	2338911	39.24	98.11
120	20	24	44	2632705	44.11	100.26

#n: Number of injections

#### Robustness

To determine the robustness of the developed method, experimental conditions were deliberately altered, and the system suitability parameter retention time and peak area were evaluated. The solution was prepared as per the test method described earlier and injected at different variable conditions like flow rate (0.8 ml/min and 1.2 ml/min.), column temperature (23 °C and 27 °C) and detection wavelength (220 nm and 222 nm). Robustness data clearly shows that the proposed method is robust at small but deliberate change [24, 25]. Robustness data are given in table 7.

Table 7: Robustness data for amitri	ptyline hydrochloride and g	zabapentin

Experiment	Amitriptyline hydr	ochloride		Gabapentin		
	<b>Retention time</b>	Average area	% assay	<b>Retention time</b>	Average area	% assay
	(min)	(n=3)		(min)	(n=3)	
Minus flow [0.8 ml/min]	4.378	2333114	98.98	2.095	390069	100.98
Plus flow [1.2 ml/min]	4.037	2331637	98.85	1.833	379152	98
Minus temperature [23 °C]	4.201	2334597	99.04	1.902	384137	99.03
Plus temperature [27 °C]	4.218	2334147	99.02	1.961	383627	98.49
Minus wavelength [220 nm]	4.221	2401311	101.1	1.961	392955	101.15
Plus wavelength [222 nm]	4.223	2331910	98.92	1.961	381593	98.22

#n: Number of injections

#### Sensitivity

The sensitivity of measurement of amitriptyline hydrochloride and gabapentin by use of the proposed method was estimated in terms of the limit of quantitation (LOQ) and limit of detection (LOD). LOQ and LOD were calculated by the use of the equations LOD =  $3.3\sigma/S$ 

and LOQ =  $10\sigma/S$  where  $\sigma$  is the standard deviation of intercepts of calibration plots and S is the average of the slopes of the corresponding calibration plot.

The LOD for amitriptyline hydrochloride and gabapentin were found to be 0.26 ppm and 41.90 ppm respectively, while LOQ for

amitriptyline hydrochloride and gabapentin were found to be 0.81 ppm and 126.97 ppm respectively.

# Solution stability

Sample solution was injected at different time interval and % assay was calculated.

The sample solution was found to be stable at room temperature if protected from light and moisture.

The solution stability of 72 h indicated that the sample solution can be used over a period of 72 h without any degradation of the solution (table 8).

Time	% assay (amitriptyline hydrochloride)	% assay (gabapentin)
Initial	99.16	99.15
After 1 d	99.14	99.11
After 2 d	99.17	99.17
After 3 d	99.15	99.16

#### Forced degradation studies

Forced degradation studies were performed to demonstrate the stability of the sample in different stressed conditions. The conditions used were acid and base hydrolysis, dry heat, oxidation, humidity and photolysis. The % assay of the amitriptyline hydrochloride and gabapentin with respect to untreated samples and % assay results obtained after treating the samples with various stress conditions had a difference which was within the acceptable limits. There was slightly higher degradation in UV, heat and humidity treated sample with respect to amitriptyline hydrochloride which indicates that it is unstable in these conditions and thus amitriptyline hydrochloride should be protected from light, moisture and direct heat, whereas it is relatively stable in acidic and basic condition. Gabapentin undergoes some amount of degradation in UV treated sample and base treated sample, slightly higher degradation when treated with hydrogen peroxide. Gabapentin is stable in acidic condition. As there was no interference observed from any components of the pharmaceutical dosage form or degrading products, hence it concludes that the developed method is stability indicating a method for simultaneous estimation of amitriptyline hydrochloride and gabapentin in the pharmaceutical dosage form. The data for forced degradation are tabulated in table 9.

#### Table 9: Data for force degradation study

Condition	Amitriptyli	Amitriptyline hydrochloride		1
	% assay	% Difference w. r. t. control	% assay	% Difference w. r. t. control
Untreated sample	99.31	NA	99.70	NA
Acid treated sample	98.31	01	99.20	0.5
Base treated sample	98.11	1.2	96.70	03
Peroxide treated sample	98.31	01	91.70	08
Heat treated sample	95.31	04	98.20	1.5
UV treated sample	88.29	11	97.70	02
Humidity treated sample	92.31	07	98.70	01

The results obtained from above set of observations prove that the method is useful in the qualitative and quantitative analysis of the drugs from the synthetic mixture and tablet formulation. Moreover, various analytical and HPLC methods for estimation of gabapentin and amitriptyline hydrochloride were reported alone and in combination with other drugs [1,2, 6-18] but as yet there is no reported HPLC analysis method for simultaneous estimation of gabapentin and amitriptyline hydrochloride combination and the novel method developed in this report is the first of its kind. This study reveals that the estimations can be done within 72 h at least, without having any detrimental effect on drug stability. The developed method is based on the use of a very economical solvent, had short chromatographic time and hence can be performed with ease.

# CONCLUSION

A validated RP-HPLC method was developed for simultaneous estimation of gabapentin and amitriptyline hydrochloride in the pharmaceutical dosage form. This novel HPLC method found to be simple, precise, accurate and had a short chromatographic time. The solution stability of 72 h indicated that the sample solution can be used over a period of 72 h without any degradation of the solution. The developed method was validated as per the ICH guidelines and the results obtained were well within the limits. Percent recovery and estimated concentration of active ingredient in pharmaceutical formulations showed that the amount of drug present is consistent with the label claim. Hence the proposed method was found to be satisfactory and can be applied for the analysis of gabapentin and amitriptyline hydrochloride in pharmaceutical dosage forms. This method can be utilized in routine quantitative and qualitative analysis of gabapentin and amitriptyline hydrochloride in pharmaceutical formulation and to study degradation pathway. These results indicate

that the proposed method may find practical applications as a qualitycontrol tool in the simultaneous analysis of the two drugs in combined dosage forms in quality-control laboratories.

# AUTHORS CONTRIBUTIONS

All the authors have contributed equally

#### **CONFLICT OF INTERESTS**

Declared none

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