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Review Article

ANALYSIS OF APPROACHES TO THE DEVELOPMENT AND VALIDATION OF THE METHODS OF ANALYSIS OF SOME ACTIVE PHARMACEUTICAL INGREDIENTS FROM THE GROUP OF CALCIUM CHANNEL BLOCKERS IN DRUGS AND BIOLOGICAL LIQUIDS

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ABSTRACT

Calcium channel blockers prevent calcium from entering cells of the heart and blood vessel walls, resulting in lower blood pressure. Calcium channel blockers, also called calcium antagonists, relax and widen blood vessels by affecting the muscle cells in the arterial walls. Physico-chemical analysis methods are increasingly being introduced into fundamental pharmaceutical research and pharmaceutical analysis practice, taking into account their high sensitivity, accuracy, specificity and expressiveness.

Analytical method development is increasingly being introduced into fundamental pharmaceutical research and pharmaceutical analysis practice, taking into account their high sensitivity, accuracy, specificity and expressiveness.

Search criteria was analytical method development for medicines from group of calcium channel blockers. Literature survey has been done in range of years 1990-2018 to make the review updated and comprehensive and to show the new approacheches to the development of the methods of analysis of calcium channel blockers. The sources were world recognized journals and key words used as filter were calcium channel blockers, amlodipine, nifedipine, verapamil, validation, method development, spectrophotometry, HPLC, UHPLC. Chromatographic methods of analysis amongst others have the greatest specificity and objectivity and allow for qualitative and quantitative determination of API in combinated dosage forms and biological fluids without prior separation of the components. We can conclude that analysts are constantly working on developing new methods of analysis and their optimization in order to save time and consumables, which also ensures the efficiency of the developed method. The main disadvantage of the described methods of API analysis can be considered long term from the beginning of chromatographic conditions that will provide high speed and high efficiency at lower pressure of the system. This reduces the amount of used mobile phase, which reduces cost analysis accordingly, while at the same time providing the necessary specificity, accuracy and reproducibility of the results of the analysis during quality control. Also, the reduction of analysis time is achieved by simplifying the conditions for sample preparation.

Keywords: Calcium channel blockers, Amlodipine, Nifedipine, Verapamil, Validation

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INTRODUCTION

Calcium channel blockers prevent calcium from entering cells of the heart and blood vessel walls, resulting in lower blood pressure. Calcium channel blockers, also called calcium antagonists, relax and widen blood vessels by affecting the muscle cells in the arterial walls.

Some calcium channel blockers have the added benefit of slowing heart rate, which can further reduce blood pressure, relieve chest pain (angina) and control an irregular heartbeat.

Some calcium channel blockers are available in short-acting and long-acting forms. Short-acting medications work quickly, but their effects last only a few hours. Long-acting medications are slowly released to provide a longer lasting effect [1].

Calcium channel blockers were first identified in the lab of German pharmacologist Albrecht Fleckenstein beginning in 1964. In 1021, Avicenna introduced the medicinal use of Taxus baccata for phytotherapy in The Canon of Medicine. He named this herbal drug "Zarnab" and used it as a cardiac remedy. This was the first known use of a calcium channel blocker drug, which were not in wide use in the Western world until the 1960s. [2].

Physico-chemical analysis methods are increasingly being introduced into fundamental pharmaceutical research and pharmaceutical analysis practice, taking into account their high sensitivity, accuracy, specificity and expressiveness.

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development of the methods of analysis calcium channel blockers. The sources were world recognized journals and key words used as filter were calcium channel blockers, amlodipine, nifedipine, verapamil, validation, method development, spectrophotometry, HPLC, UHPLC.

Methods of analysis of amlidipine

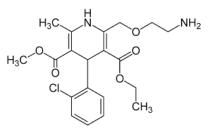


Fig. 1: The chemical structures of amlodipine

The State Pharmacopoeia of Ukraine (SPhU) does not have a monograph on the substance of amlodipine besylate or on the prepared medical form. However, the United States Pharmacopoeia regulates the determination of amlodipine besylate in substances and tablets. For identification, UV-spectrophotometry and HPLC/UV are proposed. For quantitative determination of amlodipine besylate in tablets–HPLC/UV, respectively. Chromatographic conditions for the determination of amlodipine besylate, tablets are given in the monograph of the United States Pharmacopoeia, which is used the chromatographic column of category L1 (fixed phase C18) and mobile phase consisting of three components: buffer solution of pH 3.0 with triethylamine, acetonitrile and methanol. The solvent– mobile phase, mobile phase rate–1 ml/min, detection wavelength– 237 nm. The proposed method of the United States Pharmacopeia requires a long sampling.

The European Pharmacopeia has a monograph on the substance of amlodipine besylate. Identification of amlodipine besylate of the European Pharmacopeia regulates the absorption spectro-photometry in the infrared region and quantitative determination–HPLC/UV. As a solvent, methanol is used, mobile phase–2.3 g/l of ammonium acetate solution P: methanol P (30:70, V/V), mobile phase rate–1.5 ml/min, detection wavelength–237 nm.

Methods of quantitative determination of amlodipine besylate by the method of spectrophotometry are described in the scientific literature-by products of reaction with various reagents and by their own light absorption, kinetic-spectrophotometric method in substance and drugs, by the method of inhibition of chemiluminescence, by chromatography methods.

Historical development of methods for the quantitative determination of amlodipine besylate in substances and drugs is closely related to the development of analytical methods themselves and pharmaceutical analysis in general. Nowadays the literature contains a large number of scientific papers devoted to the quantitative determination of amlodipine besylate and other APIs in one medical form, since amlodipine is used in combination with different APIs for the treatment of hypertension. Specifically, the researchers proposed spectrophotometric methods for the determination of amlodipine besylate and atorvastatin in tablets [4-6], Dake A. S. et al.-spectrophotometric method for the determination of amlodipine besylate and enalapril maleate in tablets [7], Jain H. K. et al.-spectrophotometric method for the determination of amlodipine besylate and lisinopril in tablets [8-10], Meyya S. N. et al.-spectrophotometric method for the determination of amlodipine besylate and benazepil in medical form [11-15], Wankhede S. B. et al.-spectrophotometric method for the determination of amlodipine besylate, losartan potassium and hydrochlorothiazide in tablets [16].

The HPLC method is widely used in the analysis of amlodipine besylate both in substance and in mono-and combined drugs. Simple, precise, sensitive methods are developed today–liquid chromatography, combination of TLC with densitometric determination, HPLC/UV, HPLC/DMD, HPLC/MS. For example, scientists of the National Pharmaceutical University conducted the verification of the method of reverse-phase HPLC with UV-spectrophotometric detection, which is described in the EPh. Validation characteristics such as correctness, linearity, precision, specificity, and intra-laboratory precision were studied. This HPLC/UV method can be reproduced in the laboratory and does not depend on the excipients [17-22].

Sivakumar *et al.* proposed the method of reverse-phase HPLC with UVspectrophotometric detection of amlodipine besylate. The method was proposed using a mobile phase containing the mixture of methanol, acetonitrile and 15 mmol K_2NR_4 buffer solution (pH 5.33) (10:42.08:47.92, V/V/V), mobile phase rate-1.12 ml/min [23, 24].

Argekar A. P. *et al.* proposed the combination of liquid chromatography and TLC with densitometric determination of amlodipine and atenolol in tablets using a mixture of methylene chloride, methanol, ammonia solution (8.8:1.3:0.1, V/V) as mobile phase [25-27].

Reverse-phase HPLC with UV-spectrophotometric detection method for the determination of amlodipine besylate and enalapril maleate in medical form was described by Bharat G. Chaudhari. The developed method uses the chromatographic column Phenomenex C18, 5 μ m, 250 mm×4.6 mm i.d. column and mobile phase–a mixture of methanol, acetonitrile, water (40:50:10, V/V/V). The method is linear in the range of concentrations of 0.5-6.0 μ g/ml and 0.5-8.0 μ g/ml of enalapril maleate and amlodipine besylate, respectively. The retention time for enalapril maleate and amlodipine besylate was 2.27 and 5.07 min, respectively [28, 29].

Sohan S Chitlange *et al.* developed the method of reverse-phase HPLC with UV-spectrophotometric detection of amlodipine and metoprolol in tablets using Kromasil C18 column (250 x 4.6 mm, 5 μ m) and mobile phase-a mixture of 0.02 M phosphorus buffer solution and acetonitrile (70:30 V/V, pH 3.0), mobile phase rate-1.0 ml/min, detection wavelength-221 nm [31].

The retention time for amlodipine was 2.57 min and metoprolol was 4.49 min. The linearity of the method was determined in the range of concentrations of 10-110 µg/ml of both analytes [32-34]. The same authors proposed the method of reverse-phase HPLC with UVspectrophotometric detection of the determination of amlodipine and valsartan in capsules using the Kromasil C18 column (250 x 4.6 mm, 5 µm) and mobile phase-a mixture of acetonitrile and phosphate buffer solution (0.02M, pH 3.0), (56:44 V/V), mobile phase rate-1.0 ml/min, detection wavelength-234 nm. The retention time for amlodipine was 3.07 min and valsartan was 6.20 min. Shi-Ying Dai et al. developed the method of reverse-phase HPLC with UV-spectrophotometric detection of the determination of amlodipine and ramipril in the presence of impurities in the mode of gradient elution [35]. Prathyusha W. et al. proposed the HPLC/UV method for the determination of amlodipine besylate, alixerine hemifumarate and hydrochlorothiazide in drugs using losartan as internal standard. Chromatographic conditions: Inestsil column-ODS C18 (100x4.6 mm, 5 µm), mobile phase-a mixture of 0.1 M ammonia buffer solution and acetonitrile (65:35 V/V), mobile phase rate-1.0 ml/min, detection wavelength-232 nm. The retention time for amlodipine was 5.22 min, alixerine hemifumarate was 3.90 min, and hydrochlorothiazide was 1.91 min [36-49].

Shalini Pant *et al.* described the HPLC/UV method for the determination of amlodipine besylate and bisoprolol fumarate in drugs using a C18 column (250 x 4.6 mm, 5 μ m) and mobile phase-a mixture of acetonitrile and phosphate buffer solution (50:50 V/V), mobile phase rate-1.0 ml/min, detection wavelength-230 nm. The retention time for amlodipine was 6.4 min, bisoprolol was 4.4 min. Linearity of the method was studied in the range of concentrations of 1.3-10.8 μ g/ml [50].

Reverse-phase HPLC with UV-spectrophotometric detection method for the determination of amlodipine besylate and telmisartan in the medical form was described by Kayal SD *et al.* The proposed method uses the chromatography column Prontosil C18, 5 μ m, 250 mm × 4.6 mm i.d. column and mobile phase–a mixture of methanol, potassium dihydrogen phosphate buffer solution pH 4.5 (75:25 V/V), mobile phase rate–1.4 ml/min, detection wavelength–240 nm. The method is linear in the range of concentrations of 2-20 μ g/ml and 16-160 μ g/ml amlodipine and telmisartan, respectively. The retention time for amlodipine and telmisartan was 3.12 and 5.80 min, respectively [51-55].

Raju *et al.* described the method of reversed-phase HPLC with UVspectrophotometric detection for the determination of amlodipine and perindopril in combined medical form using Xterra C18 column (100 x 4.6 mm, 5 μ m) and mobile phase-a mixture of phosphorus buffer soultion and acetonitrile (65:35 V/V), mobile phase rate-0.6 ml/min, detection wavelength-237 nm. Linearity was studied in the range of concentrations of 10-50 μ g/ml of amlodipine and 200-1000 μ g/ml of perindopril. The retention time for amlodipine and perindopril was 8.51 and 5.28 min, respectively [56-64].

The literature describes bioanalytical methods for the quantitative determination of amlodipine besylate in plasma. Bhatt J. *et al.* developed the express method for the determination of amlodipine in blood plasma using HPLC/MS and HPLC/MS/MS. The proposed method uses the chromatographic column Hypersil BDS C18, and internal standard–imipramide. The method is linear in the range of concentrations of 0.1-10 ng/ml. The total time of chromatography was 3.2 min [65].

Alsarra I. A. proposed the bioanalytical method for the determination of amlodipine in blood plasma and medical form using the developed methodology for the study of pharmacokinetics. The method was proposed using C18 Hypersil HyPurity column (3 microm, 3.9 mm id x 150 mm) and mobile phase containing a mixture of acetonitrile, potassium dihydrogen phosphate buffer solution (0.05 M) and acetic acid (62:38:0.1 V/V/V), mobile phase rate–1.8 ml/min. LOD of amlodipine besylate–1.0 ng/ml, limit of quantitation (LOQ)–10.0 ng/ml. The developed method was used for the analysis of amlodipine tablets and for the study of pharmacokinetics [66].

Feng Y. *et al.* described the HPLC/MS method for the determination of amlodipine in blood plasma. After extraction with ethyl acetate using the internal standard of nicardin, the chromatographic column C18 and mobile phase–a mixture of methanol and 1% acetic acid solution (65:35 V/V) were used [67].

Zarghi A. *et al.* developed the bioanalytical method for the determination of amlodipine in blood plasma using the described methodology for the study of pharmacokinetics. The developed procedure uses the chromatographic column Nucleosil C8, 5 μ m, 125 mm × 4.6 mm i.d. column and mobile phase–a mixture of acetonitrile and 0.01 M sodium dihydrogenphosphate buffer solution (37:63, V/V), mobile phase rate–1.5 ml/min, detection wavelength–239 nm. Linearity of the method was studied in the concentration range of 0.5-16 ng/ml. LOQ of amlodipine besylate–0.2 ng/ml [68].

Bahrami G. *et al.* proposed the express HPLC method for the determination of amlodipine in serum with fluorescence detection and application of the developed method for pharmacokinetic studies. Amlodipine was extracted from the blood serum of ethyl acetate. Pre-colon derivatization with 4-chloro-7-nitrobenzofuran was carried out. Chromatography was performed using the C18 column and mobile phase consisting of phosphate buffer solution pH 2.5 (1 ml/l of triethylamine and methanol), mobile phase rate-2.8 ml/min. As internal standard, propranolol was used. The method of quantitative determination of amlodipine besylate in blood serum was 0.25 ng/ml. The developed method found its application for the study of bioequivalence of tablets of amlodipine 10 mg [69, 70].

Scientists Y. Ma, F. Qin, X. Sun studied the pharmacokinetics of amlodipine using UHPLC/MS. Chromatography was performed using the chromatographic column ACQUITY UPLC BEH C (18) column (50 mm x 2.1 mm, id, 1.7 microm) and gradient elution (mobile phase-water and acetonitrile) with mobile phase rate-0.35 ml/min, internal standard-nimodipine The method was linear in the range of 0.15-16.0 ng/ml with lower limit of quantitation (LLOQ)-0.15 ng/ml. The described method was used to study pharmacokinetics in clinical studies [71, 72].

Scientists Q. Yu, Z. Y. Hu, F. Y. Zhu developed HPLC/MS method for determination of amlodipine and atorvastatin in blood plasma. The method was proposed using a C18 chromatographic column and gradient elution of mobile phase containing a 0.1% solution of musk acid in water and 0.1% of solution of musk acid in acetonitrile, mobile phase rate–0.4 ml/min. The developed method was used for the determination of amlodipine and atorvastatin in joint application for the treatment of patients with hypertension [73].

Wei X. *et al.* proposed the bioanalytical method for the determination of amlodipine and nicardin in blood plasma using a monolithic cation exchange column. The developed method was linear in the range of concentrations of 0.5-50.0 ng/ml of both analytes, correlation coefficient–0.998, LOQ–0.2 ng/ml of amlodipine and nicardin [74].

Scientists Ramani A. V., Sengupta P., Mullangi R. developed the UHPLC/MS method for the determination of amlodipine, simvastatin and valsartan in blood plasma with the use of pharmacokinetics. The method was proposed using the chromatographic column X-Terra C18 column and mobile phase containing a mixture of 0.02 M ammonium (pH 4.5) and acetonitrile (20:80 V/V), mobile phase rate-0.5 ml/min. The method is linear in the range of 0.5-50 ng/ml of valsartan and 0.2-50 ng/ml of amlodipine and simvastatin. The correlation coefficient was more than 0.996 for all analytes. The developed method was used for pharmacokinetic studies [75-97].

Methods of analysis of nifedipine

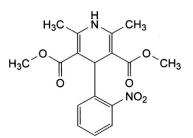


Fig. 2: The chemical structures of nifedipine

Nifedipine is a dihydropyridine calcium channel blocking agent. Nifedipine inhibits influx the transmembrane of extracellular calcium ions into myocardial and vascular smooth muscle cells, causing dilatation of the main coronary and systemic arteries and decreasing myocardial contractility. Chemical name of nifedipine is dimethyl 2,6-dimethyl-4-(2-nitrophenyl)-1,4dihydropyridine-3,5-dicarboxylate (fig. 2) [3].

The State Pharmacopoeia of Ukraine (SPhU) has a monograph on the substance of nifedipine. For identification of the substance of nifedipine, the SPhU offers the determination of melting point, absorption spectrophotometry in the infrared region, TLC (mobile phase-ethyl acetate P-cyclohexane (40:60)) and qualitative reaction of formation of azo dye after the previous restoration of nitro group to amino group, quantitative determination-cerimetry. The United States Pharmacopoeia regulates the definition of nifedipine in substances and identification, the definition of absorption tablets. For spectrophotometry in the infrared region and UV-spectrophotometry is proposed. For quantitative determination of nifedipine in tablets-HPLC/UV method. In accordance with this article, the following chromatographic conditions are used: chromatographic column of category L1 (with fixed phase C18) with sixe 4.6 mm x 250 mm; mobile phase-acetonitrile: methanol: water (25:25:50); solventmethanol, wavelength-235 nm, flow rate-1.0 ml/min.

The European Pharmacopoeia has a monograph on the substance of nifedipine. For identification, it is proposed to determine the melting point, absorption spectrophotometry in the infrared region, TLC (mobile phase-a mixture of ethyl acetate P and cyclohexane P (40:60 V/V) and qualitative reaction to the primary aromatic amino group-reaction of formation of azo dye (after preliminary reduction of nitro group to amino group). For the quantitative determination of nifedipine-method of cerimetry.

Methods of quantitative determination of nifedipine by methods of spectrophotometry and chromatography are described in the scientific literature.

Scientists Sathis Kumar Dinakaran, Babitha Alluri, Koushik Reddy Annareddy developed the spectrophotometric method for the determination of nifedipine and atorvastatin in substances and medical form. For the analysis of nifedipine and atorvastatin, it is recommended to use the UV-spectrophotometry method in methanol, whose solution has maxima at wavelengths of 237 nm (atorvastatin) and 297 nm (nifedipine) [98].

Revathi R. *et al.* proposed spectroscopic determination of nifedipine using 40% sodium salicylate solution as hydrotropic solubilizing agent. The solutions of complex have absorption maxima at 350 nm and are subject to the Bouguer-Lambert-Beer's law in the range of concentrations of 20-100 μ g/ml. The proposed method was successfully applied to determine APIs in tablets [99, 100].

Spectrophotometric determination of nifedipine in drugs, blood serum and urine using oxidation-reduction reactions was developed by scientists Tulasamma P. and Venkateswarlu P. The proposed methods are based on the restoration of the nitro group to the amino group. Method A is based on oxidation followed by the combination of nifedipine and 3-methyl-2-benzothiazolinone hydrazone in the presence of iron chloride to form a green chromogen color at 685 nm. Method B is based on the oxidation reaction of nifedipine brucine to form a purple chromogen at 546 nm [101].

Modi Dixita V., Patel Paresh U. developed the method for determination of nifedipine and telmisartan in mixtures. As a solvent, methanol was used. The maximum absorption of nifedipine-235 nm and telmisartan-297 nm. The method was linear in the range of 2-20 μ g/ml of nifedipine and 1-18 μ g/ml of telmisartan. The developed method can be used in routine analysis of drugs [102].

Scientists AlGhannam S. M., AlOlyan A. M. proposed the spectrofluorimetric method for the determination of certain substances from the group of 1,4-dihydropyridine, namely nicardipine, nifedipine, isradipine in drugs and biological fluids. The method is based on the restoration of nicardipine, nifedipine, isradipine from Zn/HCl, and fluorescence measurement ($\lambda_{em}/\lambda_{ex}$) at 460/364, 450/393 and 446/360 nm, respectively. The method was linear in the range of concentrations of 0.4-6.0, 0.2-4.0 and 0.1-9.0 µg/ml with LOD 0.0028, 0.017 and 0.016 µg/ml, respectively. The proposed method was successfully applied for the determination of nicardipine, nifedipine, isradipine in medical form and biologically advanced liquids (plasma, urine) [103].

Scientists Bing L., Hu D. F., Liu F. developed the HPLC/UV method for the determination of nifedipine and atenolol in tablets. According to the proposed method, the following chromatographic conditions are used: chromatographic column Phenomenex-ODS 3 column (250 mm×4.60 mm, 5μ m); mobile phase –methanol: water: phosphate buffer solution; wavelength–240 nm, flow rate–1.0 ml/min. The linear dependence is maintained at intervals of 10-250 µg/ml of atenolol (correlation coefficient 0.9999) and 4-100 µg/ml of nifedipine (correlation coefficient 1.0000) [104].

Scientists Cristobal Galan-Rodriguez, Jaime González-Álvarez, Màrius Valls-Remolí proposed the UHPLC/UV for the determination of nifedipine and its impurities. The chromatography column Acquity Shield RP₁₈ (50×3.0 mm, 1.7µm) and mobile phase -a mixture of 10 mmol ammonium (pH 4.5) and methanol, were used at flow rate of 0.5 ml/min. The total time of chromatography was 11 min. The method was linear in the range of 0.25-1.5 µg/ml of nifedipine and its impurities. LOQ-0.05 µg/ml [105, 106].

Rosseel M. T. *et al.* developed the method for determination of nifedipine in blood plasma using capillary gas chromatography with nitrogen detection. Nifedipine was extracted from plasma with toluene. Nitrendipine was used as an internal standard. The method was linear in the range of 2-300 ng/ml [107].

Scientists Soons P. A., Schellens J. H., Roosemalen M. C., Breimer D. D. described the analysis of nifedipine and its pyridine metabolite of dihydronifedipine in blood plasma using the method of reverse-phase HPLC/UV. The published method in 1991 was observational and of recommendatory nature [108].

Guo Y. *et al.* proposed the bioanalytical method for the determination of nifedipine in plasma and its application for the study of bioequivalence. HPLC/MS method for the determination of nifedipine in blood plasma was developed. Nifedipine was extracted with diethyl ether, dimethoxane was used as an internal standard. The chromatography column BDS C (18) and mobile phase-a mixture of methanol and water (66:34, V/V) were used. Researches on the study of bioequivalence were conducted using the developed method of 20 volunteers [109].

Wang D., Jiang K., Yang Sh. developed the bioanalytical method for the determination of nifedipine in blood plasma by the method of UHPLC/MS using the developed methodology for the study of pharmacokinetics. The chromatographic determination was performed using the chromatography column UPLCTM BEH C (18) column and mobile phase-mixture of acetonitrile and 10 mmol/l of ammonium acetate solution (75:25, V/V) at flow rate of 0.2 ml/min. The method was linear in the range of concentrations of 0.104-52.0 ng/ml, LOQ-0.104 ng/ml. The method was fully validated and successfully applied to study the pharmacokinetics of nifedipine tablets [110].

Le Guellec C. *et al.* described the definition of nifedipine in blood plasma by the method of gas chromatography. The developed

method was linear in the concentration range of 0.5-500 ng/ml, LOQ-0.5 ng/ml. The method was used to monitor the concentration of nifedipine in patients receiving this drug [111].

Zendelovska D. et al. proposed the bioanalytical method for determination of nifedipine in blood plasma. Chromatography was carried out using the mobile phase–a mixture of 0.02 mol/l potassium dihydrogen phosphate (pH 4.8) and acetonitrile (42:58, V/V) at wavelength 240 nm. Linearity of method was determined in the range of concentrations 5.0-200.0 ng/ml, LOQ–5.0 ng/ml [112, 113].

Scientists Gurley B. J., Buice R. G., Sidhu P. developed the reverse-phase HPLC/UV method for determination of nifedipine in plasma using an internal standard of 17- α -ethinylestradiol, mobile phase-a mixture of phosphate buffered solution (pH 6.1), methanol, acetonitrile (20:35:45 V/V/V), chromatography column muBondapak C-18 column at wavelength of 235 nm and flow rate of 1.0 ml/min [114, 115].

Abou-Auda H. S. *et al.* described the bioanalytical method for the determination of nifedipine in plasma and application of the developed method for the study of pharmacokinetics. Chromatography was performed using a C18 column, mobile phase-a mixture of acetonitrile, methanol and water (35:17:48, V/V/V) and internal standard-diazepam. The method was linear in the concentration range of 10-200 ng/ml and was successfully applied for pharmacokinetic studies [116].

Scientists Mosab A., Zahaa A., Momir M. proposed the HPLC/UV method for determination of nifedipine in plasma of rats to study pharmacokinetics and bioequivalence. In this method, the chromatographic column Phenomenex Luna-C18 (250 x 4.6 mm, 5 μ m) and mobile phase-a mixture of ammonium, methanol, acetonitrile (55: 43: 2, V/V/V), flow rate-0.8 ml/min, wavelength-235 nm. Linearity of method was determined in the range of concentrations 5.00-200 ng/ml, LOQ-6 ng/ml [117].

Scientists Yritia M., Parra P., Iglesias E., Barbanoj J. M. developed the HPLC/UV method for determination of nifedipine in blood plasma using the chromatographic column C18 and mobile phase–a mixture of acetonitrile and 13 mmol phosphate buffer solution pH 7 (65:35, V/V) at wavelength of 338 nm. LOD–2 ng/ml. The developed method was successfully applied for bioequivalence [118].

Wang X. D. et al. described the HPLC/MS method for the determination of nifedipine and dihydronifedipine in blood plasma, followed by its use for the study of clinical interaction of drugs. Nitrendipine was used as an internal standard. Chromatographic column–Hypersil BDS C (18) column (50 mm x 2.1 mm, i.d., 3 microm). Total chromatography time was 2.5 min. The method was linear in the range of concentrations of 0.5-100 ng/ml of both analytes. LOQ of nifedipine and dihydronifedipine–0.5 ng/ml. The developed method was successfully applied for pharmacokinetic studies [119].

Other scientists also developed the method for determination of nifedipine and dihydronifedipine in blood plasma using the chromatographic column C18 and mobile phase-a mixture of methanol and 50 mmol ammonium acetate solution (50:50, V/V). The method was linear in the range of concentrations of 0.5-100 ng/ml of both analytes [120-121].

Methods of analysis of verapamil

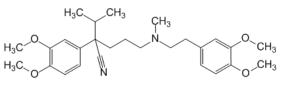


Fig. 3: The chemical structures of verapamil

Verapamil is a phenylalkylamine calcium channel blocking agent. Verapamil inhibits the transmembrane influx of extracellular calcium ions into myocardial and vascular smooth muscle cells, causing dilatation of the main coronary and systemic arteries and decreasing myocardial contractility. Chemical name of verapamil is 2-(3,4-dimethoxyphenyl)-5-[2-(3,4-dimethoxyphenyl)ethyl-methylamino]-2-propan-2-ylpentanenitrile (fig. 3) [3].

The State Pharmacopoeia of Ukraine (SPhU) has a monograph on the substance of verapamil hydrochloride and on verapamil hydrochloride tablets. To identify the substance of verapamil hydrochloride, SPhU offers UV-spectrophotometry, absorption spectrophotometry in the infrared region, TLC (mobile phase-diethylamine P-cyclohexane (15:85)), quantitative determination-alkalimetry, potentiometric titration. For identification of verapamil hydrochloride in tablets, SPhU proposes UV-spectrophotometry, HPLC/UV (mobile phase-a mixture of heptylamine P^N-acetic acid of ice P-acetonetrile P-solution of 1.36 g/l of sodium acetate P (1:4.7:58:1.37)), qualitative reaction to chlorides. For quantitative determination of verapamil hydrochloride in tablets–UV-spectrophotometry.

The United States Pharmacopoeia regulates the definition of verapamil hydrochloride in substance, tablets and injection solution. In order to identify verapamil hydrochloride in the substance, determination is made by absorption spectrophotometry in the infrared region, HPLC/UV (mobile phase-analogous to SPhU), qualitative reaction to chlorides, for quantitative determinationacidimetry in non-aqueous medium. For the identification of verapamil hydrochloride in tablets, the United States Pharmacopoeia offers the absorption spectrophotometer method in the infrared area and HPLC/UV. The drug Verapamil hydrochloride, tablets are described in the monograph of the United States Pharmacopoeia. According to this article, the HPLC/UV is regulated by the following chromatographic conditions: chromatographic column of category L1 (fixed phase C18) size 4.6 mm x 125 or 150 mm; mobile phaseacetonitrile: 2-aminoheptane: solution A (0.015 M solution of sodium acetate containing 33 ml/l of acetic acid) in ratio (30:0.5:70); wavelength-278 nm, flow rate-0.9 ml/min.

The European Pharmacopoeia [3] suggests identifying verapamil hydrochloride UV-spectrophotometry substance, absorption spectrophotometry in the infrared region, TLC (mobile phase-diethylamine P-cyclohexane (15:85)) and qualitative reaction to chlorides, quantitative determination-alkalimetry potentiometric titration.

Methods of quantitative determination of verapamil hydrochloride by spectrophotometry, electrochemical method and chromategraphy methods are described in the scientific literature.

Salh D. M. proposed spectrophotometric determination of verapamil in drugs using bromotymol blue. The method is based on the reaction of complex formation. The reaction product was extracted with tetrachloromethane and the optical density was measured at 420 nm. The developed method was completely validated. Coefficient of co-correlation–0.9951, LOQ–0.009253 μ g/ml. The method was successfully applied for the analysis of medical form containing verapamil hydrochloride (tablets, capsules, ampoules) [122].

Scientists Rahman N., Hoda M. N. developed the spectrophotometric method for the determination of verapamil hydrochloride in drugs using chloramine-T. The method is based on the oxidation reaction of verapamil with chloramine-T in acidic environment. The solutions of the complex are yellow in color and have absorption maxima at 425 nm and are subject to the Bouguer-Lambert-Beer's law in the range of concentrations of 0-340 μ g/ml. The proposed method was successfully applied to determine verapamil hydrochloride in medical form [123].

Scientists Nafisur Rahman and Syed Najmul Hejaz Azmi proposed the spectrophotometric method of verapamil hydrochloride in drugs using *N*-bromosicinamide. The method is based on the oxidation of verapamil hydrochloride by *N*-bromosicinamide in chlorinated acid environment at room temperature to give a yellow colored product having maximum absorption at 415 nm and subject to the Bouguer-Lambert-Beer's law for the concentration range 10.0-200.0 μ g/ml. The developed method was successfully applied for the determination of verapamil hydrochloride in the drug [124].

Nafisur Rahman *et al.* described two spectrophotometric methods for the determination of verapamil hydrochloride in drugs. Method

A is based on the oxidation of verapamil potassium hydrochloride with metaperiodate in environment of sulphate acid to the formation of colored reaction product having a maximum absorption at 425 nm and subject to the Bouguer-Lambert-Beer's Law in the range of 12.5-187.5 μ g/ml. Method B is based on the formation of colored reaction product with tropaeolin 00 at pH 4.0, which is excreted by chloroform, has a maximum absorption at 400 nm and is subject to the Bouguer-Lambert-Beer's law in the range of 2.0-30.0 μ g/ml. Both methods have found their application in the analysis of drugs containing verapamil hydrochloride [125-130].

Ivanova V. *et al.* developed the HPLC/UV method for determination of verapamil in blood plasma using the chromatographic column Lichrospher 60 RP-select B column (250 mm × 4 mm ID, 5μm) and mobile phase-a mixture of 40% acetonitrile and 0.025 mol/l of potassium dihydrogenphosphate pH 2.5, flow rate-1 ml/min, detection wavelength-200 nm. The method was linear in the range of 10-500 ng/ml. The developed method was successfully applied for routine analysis of verapamil in blood plasma [131-133].

Sawicki W. proposed the validated method for determination of verapamil and its metabolite norverpamil in blood plasma. LOD of verapamil is 0.924 ng/ml, LOQ-3.080 ng/ml. LOD of norvermapamil-0.030 ng/ml, LOQ-1.001 ng/ml. The total time of chromatography is 8 min, which makes it possible to make 30-40 manual analyses per day (without autosampler). The developed method was successfully applied for carrying out of pharmacokinetic studies and researches on studying bioequivalence of drugs containing verapamil [134, 135].

Scientists Stagni G. and Gillespie W. R. described the method for the determination of verapamil and its norverapamil metabolite in blood plasma. Norverapamil is rapidly acetylated to *N*-acetylnorverapamil using 2% butanol in hexane. Verapamil and *N*-acetylnorverapamil are separated by α -glycoprotein chiral column by mobile phase-a mixture of phosphate buffered solution (0.01 M, pH 6.65) and acetonitrile. The wavelengths of the fluorescence detector were set at 227 nm for excitation and 308 nm for radiation. The method was linear in the range of concentrations of 3-200 ng/ml of verapamil and 2-100 ng/ml of norverapamil. LOD of verapamil was 3 ng/ml and norverapamil-2 ng/ml. The developed method was used to study pharmacodynamics [136].

Jhee OH *et al.* developed the bioanalytical method for determining verapamil in plasma of rats using mobile phase A (50 mmol ammonium phosphate, pH 4.5) and mobile phase B (50 mmol ammonium phosphate: acetonitrile, 70:30 v/v). Analysis of verapamil and internal standard of propranolol were performed by direct introduction of rat plasma into the system and eluted, respectively, 22 min (verapamil) and 12 min (propranolol) at mobile phase A of 0.5 ml/min and mobile phase B of 0.15 ml/min. The method was validated. The correlation coefficient was 0.9997. The method was linear in the range of 0.01 to 2.50 µg/ml. LOD-0.01 µg/ml, LOQ-0.025 µg/ml. The developed method was successfully used for the study of pharmacokinetics [137, 138].

The method of reverse HPLC/UV determination of verapamil in blood plasma was developed by Muscara M. N. and de-Nucci G. Flecainide was used as internal standard. The method was completely validated and used for pharmacokinetic studies [139].

Scientists Lau-Cam C. A. and Piemontese D. proposed the HPLC/UV method for determining verapamil in plasma of rats using dextromethorphan as an internal standard and mobile phase-a mixture of methanol, acetonitrile, triethylamine acetate buffer solution (10:30:60 V/V/V) with detection wavelength of 235 nm, flow rate of 1.5 ml/min. The method was linear in the concentration range of 0.5-10 µg/ml (correlation coefficient-0.9999). LOD-0.1 µg/ml. The method was used to monitor the concentration of verapamil in the plasma of rats receiving this drug through various routes of administration [140].

Scientists Hedeland M., Fredriksson E., Lennernäs H., Bondesson U. described the bioanalytical method for the determination of verapamil enantiomers and their *N*-dimethylated metabolite in blood plasma using HPLC/MS. The mobile phase–a mixture of 85% aqueous solution of ammonium acetate pH 7.4 and a 15% solution of

acetonitrile, flow rate-0.6 ml/min. The developed method was successfully applied for the study of *in vivo* intestinal permeability [141-144].

CONCLUSION

From the above-mentioned, we can conclude that analysts are constantly working on developing new methods of analysis and their optimization in order to save time and consumables, which also ensures the efficiency of the developed method. The main disadvantage of the described methods of API analysis can be considered long term from the beginning of chromatography to API release and specific solvents used as the mobile phase in HPLC. In our opinion, it is necessary to develop methods and to select such chromatographic conditions that will provide high speed and high efficiency at lower pressure of the system. This reduces the amount of used mobile phase, which reduces cost analysis accordingly, while at the same time providing the necessary specificity, accuracy and reproducibility of the results of the analysis during quality control. Also, the reduction of analysis time is achieved by simplifying the conditions for sample preparation.

AUTHORS CONTRIBUTIONS

All the author have contributed equally

CONFLICT OF INTERESTS

Declared none

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