

COMPARATIVE STUDIES OF NOVEL OXADIAZOLE DERIVATIVE HAVING CHIRAL CENTER AND THEIR ANTI-MICROBIAL ACTIVITIES

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ABSTRACT

Objective: To synthesize a series of (-) S & (+) S oxadiazole derivatives by the reaction of (-) (2S)-2-amino-2-(2-chlorophenyl) acetohydrazide & (+) (2S)-2-amino-2-(2-chloro phenyl) acetohydrazide with con. H₂SO₄, L(+) Tartaric acid, L(-) Tartaric acid, Hydrazine Hydrate, POCl₃ and Various aromatic acid respectively.

Methods: The structures of novel synthesized compounds have been established on the basis of elemental analysis, IR, ¹H NMR, ¹³C NMR and mass spectral data and screened for their antibacterial and antifungal activities against different microorganisms by micro dilution method. Gentamycin, Ampicillin, Chloramphenicol, Ciprofloxacin, Norfloxacin, Nystatin, Griseofulvin were used as standard drugs for bacterial and fungus.

Results: The newly synthesized compounds indicate that some of them show better antibacterial and antifungal activity than compared to their reference drug.

Conclusion: Comparison of 1, 3, 4 oxadiazole having chiral center, characterized and exhibited promising antibacterial and antifungal activity.

Keywords: 1, 3, 4-Oxadiazol, Spectroscopic study, Antibacterial activity, Antifungal activity.

INTRODUCTION

Chirality of drugs, particularly the comparison of efficacy of enantiomers and their racemic mixtures, has become an objective of serious interest of pharmaceutical researchers. Advances in chemical technologies connected with the synthesis, separation, and analysis of pure enantiomers from racemates, together with administrative regulatory measures, have resulted in an increase in the number of newly registered chiral drugs containing only one of the enantiomers.

Polyheterocyclic compounds having an oxadiazole fragment are potential biologically active compounds. Among isomeric oxadiazoles, 1, 2, 3- and 1, 2, 4-oxadiazoles were studied in sufficient detail. Interest in 1,2,5-oxadiazole derivatives as biologically active substances has arisen relatively recently, and some representatives of this series have been found, which exhibit a wide spectrum of biological properties: antimicrobial [1], antitubercular [2-3], spasmolytic, and muscle relaxant activity etc.

1, 3, 4-Oxadiazoles [4-5] have attracted interest in medicinal chemistry as surrogates of carboxylic acids, esters and carboxamides. They are an important class of heterocyclic compounds that have a wide range of pharmaceutical and biological activities including antimicrobial [6-7], antifungal [8-9], anti-inflammatory [10], and antihypertensive. Several methods have been reported in the literature for the synthesis of 1, 3, 4-oxadiazoles.

These protocols are multi-step in nature. The most general method involves the cyclization of diacylhydrazides with a variety of reagents, such as thionyl chloride, phosphorus oxychloride or sulfuric acid usually under harsh reaction conditions. Few reliable and operationally simple examples have been reported for the one-step synthesis of 1, 3, 4-oxadiazoles, especially from readily available carboxylic acids and acid hydrazides [11].

MATERIALS AND METHODS

Melting points were determined by open capillaries and are uncorrected. The homogeneity of the compounds was checked by TLC (silica gel H, BDH, Toluene: Methanol 8:2). IR-spectra (cm⁻¹) were recorded on a Shimadzu FT-IR spectrophotometer using KBr

pellet method. ¹H NMR and ¹³C NMR spectra were recorded on a Bruker Avance 300 MHz & 75 MHz NMR instrument, using DMSO-*d*₆ as solvent and TMS as internal reference (chemical shifts in δ, ppm).

Mass spectra were obtained on a JMS-T100LC, Accu TOF Mass spectrometer (DART). The elemental analysis (C, H, N) of compounds was carried out on a Carlo Erba 1108 elemental analyzer. Their results were found to be in good agreement with the calculated values.

Experimental procedure

Synthesis of Methyl amino (2-chlorophenyl) acetate (2)

A mixture of amino (2-chlorophenyl) acetic acid (0.54 mol) and Methanol was taken in a round bottom flask. con. H₂SO₄ (1 mol) added slowly below 35°C. After complete addition the reaction mixture was refluxed for 16 hrs and the progress of the reaction was monitored by TLC (Mobile Phase Toluene: Methanol 8:2). Then methanol was distilled out completely & then toluene & water were added. After separation of two layers aq. Layer was taken & cooled to 10-15°C. Then dichloromethane was added and pH was adjusted at 7.0 to 7.5 with liq. NH₃. From the layer separated, aq. layer was taken & washed with dichloromethane. Both organic layer were taken & washed with water. Organic layer dried with Na₂SO₄ and dichloromethane distilled out completely to obtained oily residue. This ester was directly used for the second stage without carrying for any further purification. The ester was in the form of light yellow color liquid, Yield 85 %.

Synthesis of Methyl (-) (2S)-amino (2-chlorophenyl) acetate (tartrate salt) (3)

A mixture of L (-) Tartaric acid (0.36 mol) and methanol was taken in a round bottom flask. Reaction mixture was cooled at 15-20°C and methyl amino (2-chlorophenyl) acetate (0.50 mol) and acetone were added into the reaction mass at 15-20°C for 2 hrs.

After addition temperature was raised to 30-35°C which was maintained for 18 hrs. The reaction mixture was cooled at 0-5°C and maintained for 2 hrs. The reaction mixture was filtered, washed with methanol to give a white colored solid. [α]_D²⁰ + 85° to + 95° (c=1 in methanol), Melting Range 164 °C to 168 °C.

Synthesis of Methyl (+) (2S)-amino (2-chlorophenyl) acetate (tartrate salt) (4)

A mixture of L (+) Tartaric acid (0.36 mol) and methanol was taken in a round bottom flask. Reaction mixture was cooled at 15-20°C then methyl amino (2-chlorophenyl) acetate (0.50 mol) and acetone were added into the reaction mass at 15-20°C for 2 hrs. After addition temperature was raised to 30-35°C which was maintained for 18 hrs.

The reaction mixture were cooled to 0-5°C and maintained for 2 hrs. It was then filtered & washed with methanol. White color product, Yield 130 %. $[\alpha]_D^{20}$ - 85° to - 95° (c=1 in methanol), Melting Range 158 °C to 163 °C.

Free Base from tartrate salt (5)

A mixture of methyl (-) (2S)-amino (2-chlorophenyl) acetate tartrate salt, water and dichloromethane were taken into a round bottom flask pH was adjusted to 7.0 - 8.0 with liq. NH₃. From the layer separated, aqueous layer was taken & washed with dichloromethane. Both Organic layers were taken and dry with Na₂SO₄ from which dichloromethane was distilled out completely to obtained oily residue. $[\alpha]_D^{20}$ + 130° to + 135° (c = 1 in methanol), Yield 52%.

Free Base from tartrate salt (6)

A mixture of methyl (+) (2S)-amino (2-chlorophenyl) acetate tartrate salt, water and dichloromethane were taken into a round bottom flask pH was adjusted to 7.0 - 8.0 with liq. NH₃. From the layer separated, aqueous layer was taken & washed with dichloromethane. Both Organic layers were taken and dry with Na₂SO₄ from which dichloromethane was distilled out completely to obtained oily residue. $[\alpha]_D^{20}$ - 130° to - 135° (c = 1 in methanol), Yield 52%.

Synthesis of (-) (2S)-2-amino-2-(2-chlorophenyl) acetohydrazide (7)

A mixture of (-) (2S)-amino (2-chlorophenyl) acetate (1 mol) and methanol were taken in a round bottom flask. The reaction mixture was heated up to 50°C to 55°C. Hydrazine Hydrate (1.7mol) was added slowly and methanol was used as a solvents. The reaction mixture was refluxed on water bath for 5-6 hrs.

The completion of the reaction was judged by TLC (Mobile Phase Toluene: Methanol 8:2). After completion of the reaction 80% of the solvent was distilled out and the reaction mixture was cooled at 5°C -10°C. The solid thus separated was collected by filtration & dried. Off white color product, $[\alpha]_D^{20}$ + 84° to + 90° (c = 1 in methanol), Melting Range 90°C - 96°C, Yield 75 %.

Synthesis of (+) (2S)-2-amino-2-(2-chlorophenyl) acetohydrazide (8)

A mixture of (+) (2S)-amino (2-chlorophenyl) acetate (1 mol) and methanol were taken in a round bottom flask. The reaction mixture was heated up to 50°C to 55°C. Hydrazine Hydrate (1.7mol) was added slowly and methanol was used as a solvents. The reaction mixture was refluxed on water bath for 5-6 hrs. The completion of the reaction was judged by TLC (Mobile Phase Toluene: Methanol 8:2). After completion of the reaction 80% of the solvent was distilled out and the reaction mixture was cooled at 5°C -10°C. The solid thus separated was collected by filtration & dried. Off white color product, $[\alpha]_D^{20}$ - 84° to - 90° (c = 1 in methanol), Melting Range 94°C - 98°C, Yield 75 %.

Synthesis of (-)(S)-1-(2-chlorophenyl)-1-(5-phenyl-1, 3, 4-oxadiazol-2-yl) methanamine (9a-j)

A mixture of (-) (S)-2-amino-2-(2-chlorophenyl) acetohydrazide (1 mol), various aromatic acid (1.2 mol), Phosphorus oxychloride (7 mol) and toluene was taken into a round bottom flask. The reaction mixture was heated up to 100°C and the temperature was maintained for 2-4 hrs till TLC OK (Mobile Phase Toluene: Methanol: Triethylamine 8:1.5:0.5). water insted of then water added at 0-5°C and the pH adjusted to 7.5 to 8.5 with liq. NH₃ at 0-5°C. Then it was filtered, dried & recrystallized from Methanol. (9g) IR (cm⁻¹) -Cl (717), -C-O-C (1157), -C=N (1523), -NO₂ (1346), -NH₂ (3107). (9e)-H NMR [300 MHz, δ] 1.25(d, 1H, -CH), 6.17 (d, 2H, -NH₂), 7.26-8.29 (m, 7H, ArH). (9e) ¹³C NMR [75 MHz, δ] 130.73 (C₁), 126.89 (C₂), 127.39 (C₃), 127.89 (C₄), 132.11 (C₅), 139.73 (C₆), 46.83 (C₇), 169.36 (C₈), 169.36 (C₉), 136.20 (C₁₀), 130.89 (C₁₁), 131.66 (C₁₂), 136.20 (C₁₃), 131.66 (C₁₄), 133.71 (C₁₅). (9e) M/S (m/z,relative intensity): 354.61(M⁺), 359(M+4).

Synthesis of (+) (S)-1-(2-chlorophenyl)-1-(5-phenyl-1, 3, 4-oxadiazol-2-yl) methanamine (10a-j)

A mixture of (+) (2S)-2-amino-2-(2-chlorophenyl) acetohydrazide (1 mol), various aromatic acid (1.2 mol), Phosphorus ox chloride (7 mol) and toluene was taken into a round bottom flask. The reaction mixture heat up to 100°C and maintain temperature 2-4 hrs till TLC OK (Mobile Phase Toluene: Methanol: Triethylamine 8:1.5:0.5). Added water at 0-5°C and then pH adjusted 7.5 to 8.5 with liq. NH₃ at 0-5°C. Then it was filtered, dried & recrystallized from Methanol. (10) IR (cm⁻¹) -Cl (759), -C-O-C (1207), -C=N (1583), -NH₂ (3107). (10a) ¹H NMR [300 MHz, δ] 1.23 (d, 1H, -CH), 6.37 (d, 2H, -NH₂), 7.35-7.94 (m, 9H, ArH). (10a) ¹³C NMR [75 MHz, δ] 131.41 (C₁), 128.54 (C₂), 130.06 (C₃), 134.11 (C₄), 133.16 (C₅), 132.47 (C₆), 52.25 (C₇), 168.41 (C₈), 169.37 (C₉), 137.05 (C₁₀), 127.11 (C₁₁), 126.44 (C₁₂), 128.98 (C₁₃), 126.44 (C₁₄), 127.11 (C₁₅). (10b) M/S (m/z,relative intensity): 299(M⁺), 301(M+2).

Table 1: Physical, characterization data of compound (9a-j) & (10a-j)

Comp.	Functional Group	Mol. Formula	Mol. wt.	Yield (Time/hrs)	%C Required (Found)	%H Required (Found)	%N Required (Found)
9a & 10a	-H	C ₁₅ H ₁₂ N ₃ OCl	285.72	62% (2-4)	63.05 (63.00) & (63.02)	4.23 (4.20) & (4.19)	14.71 (14.70) & (14.68)
9b & 10b	3-CH ₃	C ₁₆ H ₁₄ N ₃ OCl	299.75	83% (2-4)	64.11 (64.08) & (64.10)	4.71 (4.69) & (4.69)	14.02 (14.00) & (14.02)
9c & 10c	2-OH	C ₁₅ H ₁₂ N ₃ O ₂ Cl	301.72	78% (2-4)	59.71 (59.70) & (59.69)	4.01 (3.99) & (4.00)	13.93 (13.90) & (13.91)
9d & 10d	4-OH	C ₁₅ H ₁₂ N ₃ O ₂ Cl	301.72	70% (2-4)	59.71 (59.69) & (59.68)	4.01 (4.00) & (3.99)	13.93 (13.91) & (13.92)
9e & 10e	2,4-di-Cl	C ₁₅ H ₁₀ N ₃ OCl ₃	354.61	88% (2-4)	50.80 (50.80) & (50.79)	2.84 (2.81) & (2.82)	11.85 (11.82) & (11.84)
9f & 10f	2-NO ₂	C ₁₅ H ₁₁ N ₄ O ₃ Cl	330.72	80% (2-4)	54.47 (54.45) & (54.46)	3.35 (3.32) & (3.30)	16.94 (16.91) & (16.93)
9g & 10g	4-NO ₂	C ₁₅ H ₁₁ N ₄ O ₃ Cl	330.72	75% (2-4)	54.47 (54.45) & (54.45)	3.35 (3.33) & (3.31)	16.94 (16.91) & (16.92)
9h & 10h	4-NH ₂	C ₁₅ H ₁₃ N ₄ OCl	300.74	58% (2-4)	59.91 (59.89) & (59.90)	4.36 (4.32) & (4.34)	18.63 (18.60) & (18.61)
9i & 10i	3-NH ₂ 2-OH	C ₁₅ H ₁₃ N ₄ O ₂ Cl	316.74	67% (2-4)	56.88 (56.85) & (56.87)	4.14 (4.12) & (4.13)	17.69 (17.66) & (17.67)
9j & 10j	3,5-di-NH ₂	C ₁₅ H ₁₄ N ₅ OCl	315.75	62% (2-4)	57.06 (57.03) & (56.06)	4.47 (4.45) & (4.47)	22.18 (22.15) & (22.17)

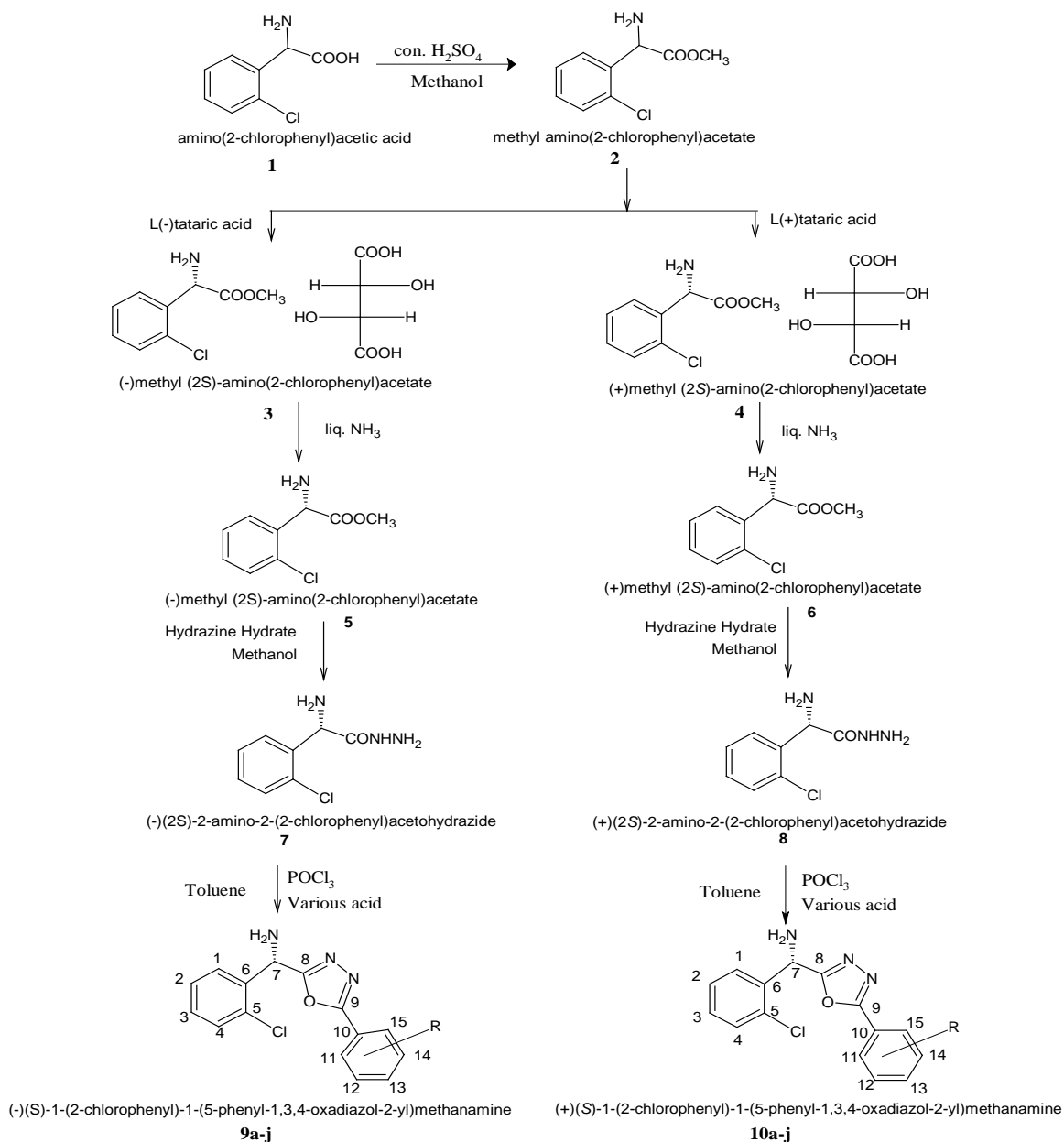


Table 2: Comparison of compound (9a-j) & (10a-j)

Comp.	Specific Optical Rotation	Melting Point	Comp.	Specific Optical Rotation	Melting Point
9a	-130°	120-126 ° C	10a	66°	227-232 ° C
9b	-50°	85-95 ° C	10b	16°	194-199 ° C
9c	-	115 ° C dec.	10c	20°	140-143 ° C
9d	-70°	135-140 ° C	10d	30°	228 ° C dec.
9e	-60°	138-144 ° C	10e	50°	173-180 ° C
9f	-80°	175-185 ° C	10f	90°	144-149 ° C
9g	-30°	118-122 ° C	10g	60°	167-178 ° C
9h	-	242-248 ° C	10h	-	188-197 ° C
9i	-	265-275 ° C	10i	-	182-186 ° C
9j	-	218-225 ° C	10j	-	80-90 ° C

Antimicrobial Activity

Following common standard strains were used for screening of antibacterial and antifungal activities: *E.Coli*, *P.Aeruginosa*, *S.Aureus*,

S.Pyogenus, *C.Albicans*, *A.Niger*, *A.Clavatus* the strains were procured from Institute of Microbial Technology, Chandigarh. DMSO was used as diluents / vehicle to get desired concentration of drugs to test upon standard bacterial strains. Each synthesized drug was diluted

for obtaining 2000 microgram /ml concentration, as a stock solution. In primary screening 1000 microgram/ml, 500 microgram /ml, and 250 microgram /ml concentrations of the synthesized drugs were taken. The active synthesized drugs found in this primary screening were further tested in a second set of dilution against all microorganisms. The drugs found active in primary screening were similarly diluted to obtain 200 microgram/ml, 100 microgram/ml, 50 microgram/ml, 25 microgram/ml, 12.5 microgram/ml and 6.250 microgram/ml concentrations. The highest dilution showing at least 99 % inhibition zone is taken as MIC. The result of this is much affected by the size of the inoculums. The test mixture should contain 10⁸ microorganism/ml. Synthesized derivatives -1,3,4-Oxadiazoles; by incorporating 2-Chlorophenyl methyl and substituted phenyl systems, at two free positions in the oxadiazole ring system were evaluated for antimicrobial activity by micro dilution method against MTCC 443 (*E.coli*), MTCC 1688 (*P.*

aeruginosa), MTCC 96 (*S. aureus*), MTCC 442 (*S. pyogenus*), MTCC 227 (*C.albicans*), MTCC 282 (*A. niger*) and MTCC 1323 (*A. clavatus*) respectively using the standard drugs Gentamycin, Ampicillin, Chloramphenicol, Ciprofloxacin, Norfloxacin, Nystatin and Griseofulvin. The Comparative activities of the newly synthesized compound (9a-j) and (10a-j) & the control antibiotics on bacterial & fungal strains respectively were summarized in Table 3.

Excellent to good activity was observed in compounds 9a, 9i, 10b, 10c & 10f (against *E.coli*), compounds 10a, 10e & 10g (against *P.aeruginosa*), compounds 9e, 9f, 9j, 10a, 10f, 10g, 10i & 10j (against *S.aureus*), compounds 9e, 9f, 10c, 10d & 10e (against *S.pyogenus*), compound 10d (against *A. niger* & *A. clavatus*) and compound 10f (against *C.albicans* & *A. niger*). The remaining compounds were found effective at a much higher concentration as compared to the standard drugs.

Table 3 Antimicrobial activity of compound (9a-j) & (10a-j)

comp.	Minimum Inhibition Concentration						
	Antibacterial				antifungal		
	E.coli MTCC443	P.aeruginosa MTCC1688	S.aureus MTCC96	S.pyogenus MTCC442	C.albicans MTCC227	A.niger MTCC282	A.clavatus MTCC1323
9a	100	100	250	250	1000	500	500
9b	125	200	500	125	1000	>1000	>1000
9c	500	500	200	200	500	500	500
9d	250	250	200	200	500	500	1000
9e	250	250	100	100	1000	1000	1000
9f	200	200	50	100	1000	1000	1000
9g	200	200	200	200	1000	1000	>1000
9h	250	250	250	250	1000	500	1000
9i	100	100	250	250	>1000	>1000	>1000
9j	250	250	125	500	>1000	>1000	>1000
10a	250	100	100	250	500	>1000	>1000
10b	50	125	500	250	1000	1000	>1000
10c	62.5	100	500	100	500	1000	1000
10d	250	200	250	100	500	250	250
10e	200	100	200	100	500	1000	250
10f	100	62.5	100	500	250	250	1000
10g	250	100	125	500	250	500	>1000
10h	200	200	250	200	1000	>1000	>1000
10i	250	250	100	500	>1000	>1000	>1000
10j	125	200	125	250	>1000	1000	500
Gentamycin	0.05	1	0.25	0.5			
Ampicillin	100	--	250	100			
Chloramphenicol	50	50	50	50			
Ciprofloxacin	25	25	50	50			
Norfloxacin	10	10	10	10			
Nystatin					100	100	100
Griseofulvin					500	100	100

RESULTS AND DISCUSSION

The compounds were synthesized as per scheme I

Amino (2-chlorophenyl) acetic acid 1 was converted to amino (2-chlorophenyl) methyl ester 2 using con.H₂SO₄ in presence of methanol at a reflux temperature. The optical isomer, (+) amino(2-chlorophenyl)methyl ester was separated from its (-) isomer using L-(+)-tartaric acid and (-) amino(2-chlorophenyl)methyl ester was separated from its (+) isomer using L-(-)-tartaric acid and the (-) tartrate salt 3 & (+) tartrate salt 4 thus obtained were converted into their free base 5 & 6 using liquor ammonia. The (-) tartrate salt & (+) tartrate salt with high enantiomeric purity were achieved by repeatedly heating and cooling the reaction mass till the required enantiomeric purity obtained. (-) (2*S*)-2-amino-2-(2-chlorophenyl) acetohydrazide 5 & (+) (2*S*)-2-amino-2-(2-chlorophenyl) acetohydrazide 6 were synthesized by reacting (-) amino (2-chlorophenyl) methyl ester & (+) amino (2-chlorophenyl) methyl ester free base react with hydrazine hydrate. The targeted (-) (S)-1-(2-chlorophenyl)-1-(5-phenyl-1,3,4-oxadiazol-2-yl) methanamine 9a-j & (+) (S)-1-(2-chlorophenyl)-1-(5-phenyl-1,3,4-oxadiazol-2-yl) methanamine 10a-j were synthesized by refluxing (-) (2*S*)-2-amino-2-

(2-chloro phenyl) acetohydrazide 7 & (+) (2*S*)-2-amino-2-(2-chlorophenyl) acetohydrazide 8 with various aromatic acid and POCl₃ as a cyclization catalyst. Compounds 9a-j and 10a-j were prepared as a new product. The proposed structures of all the synthesized compounds are well supported by elemental analysis, IR, ¹H NMR & ¹³C NMR data. Compounds 5 & 6 were insoluble in sodium bicarbonate solution indicating the involvement of acid. The ¹H NMR spectrum displayed signals for the presence of amino proton around δ 6.17 & 6.37. Aromatic protons were observed in the usual region as multiplet at δ 7.26-8.29 & 7.35-8.98.

CONCLUSION

In conclusion, we have described the synthesis and biological activities of a new 1, 3, 4-Oxadiazole derivative. The compounds **9f**, **10b**, **10c** and **10f** showed good antibacterial activity than parent Chloramphenicol. The argument is strengthened herein that stereospecificity is essential criteria towards receptor binding, particularly with respect to antibacterial activity of 1, 3, 4-Oxadiazoles. And we were successful in this first ever attempt to compare the antibacterial activity of chirally pure individual 1, 3, 4-Oxadiazole diastereomers. The structure of novel compounds was determined

by IR, ¹H NMR, ¹³C NMR and mass spectroscopic techniques and analytical methods.

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CONFLICT OF INTERESTS

Declared None

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